

Ions in Channels Natural Nanovalves

Workshop

Mathematical Models of Electrolytes in Molecular Biology

National Taiwan University

organized by

Prof. Tai Chia Lin, Chun Liu

Thank You !

and
thanks to Wei-jhen Tsai

Stochastic Derivation of PNP

done previously in some detail, starting with
Langevin Stochastic Differential Equation
using theory of Stochastic Processes,
all the way to Law of Mass Action
analytically, with evaluation of all integrals etc

Today,
**Experimental Evidence for
all spheres model**

**This is the best available test of different mathematical models,
in my opinion**

Ion Channels are the Valves of Cells

Ion Channels are the Main Controllers of Biological Function

Selectivity

Different Ions
carry
Different Signals

Chemical Bonds are lines
Surface is Electrical Potential
Red is negative (acid)
Blue is positive (basic)

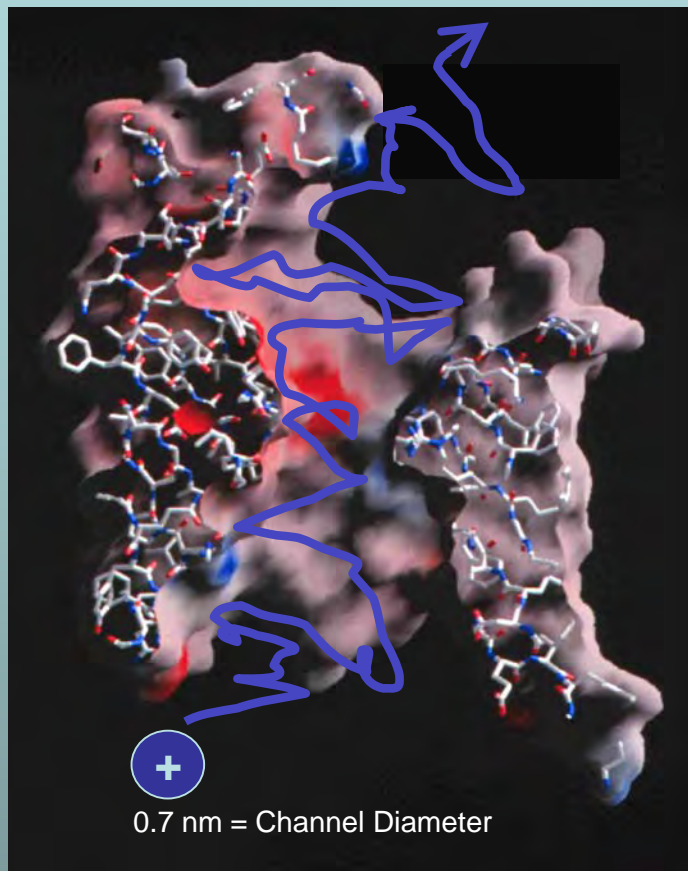


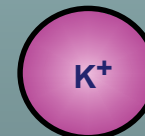
Figure of ompF porin by Raimund Dutzler

Ions in Water*

are the
Liquid of Life

*Pure H₂O is toxic to cells & proteins

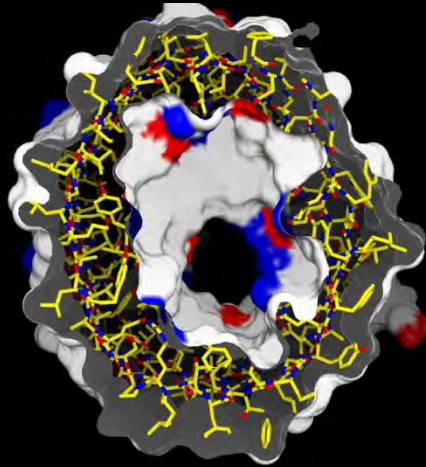
Hard Spheres



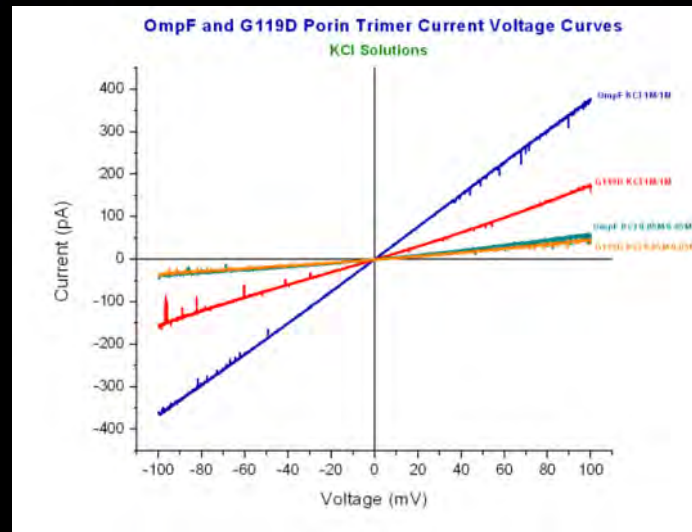
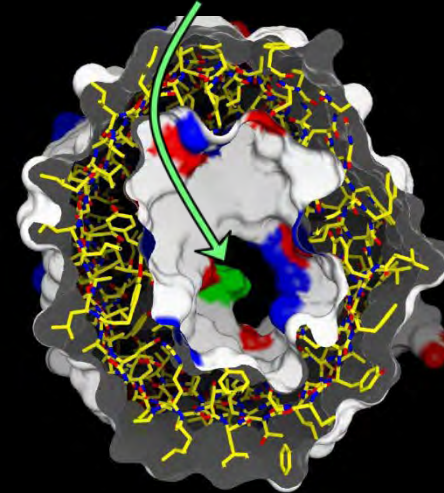
3 Å

A few atoms make a BIG Difference

Ompf



G119D



Glycine
replaced by
Aspartate

Structure determined by
Raimund Dutzler
in Tilman Schirmer's lab

Current Voltage relation
by
John Tang
in Bob Eisenberg's Lab

General Theme

Mathematics of Molecular Biology

is (mostly)

Reverse Engineering

i.e., solving specific

Inverse Problems

How does it work?

How do a few atoms control

(macroscopic) **Biological Function**

Ion Channels are Biological Devices*

Natural nano-valves** for atomic control of biological function

Ion channels coordinate contraction of cardiac muscle, allowing the heart to function as a pump

Coordinate contraction in skeletal muscle

Control all electrical activity in cells

Produce signals of the nervous system

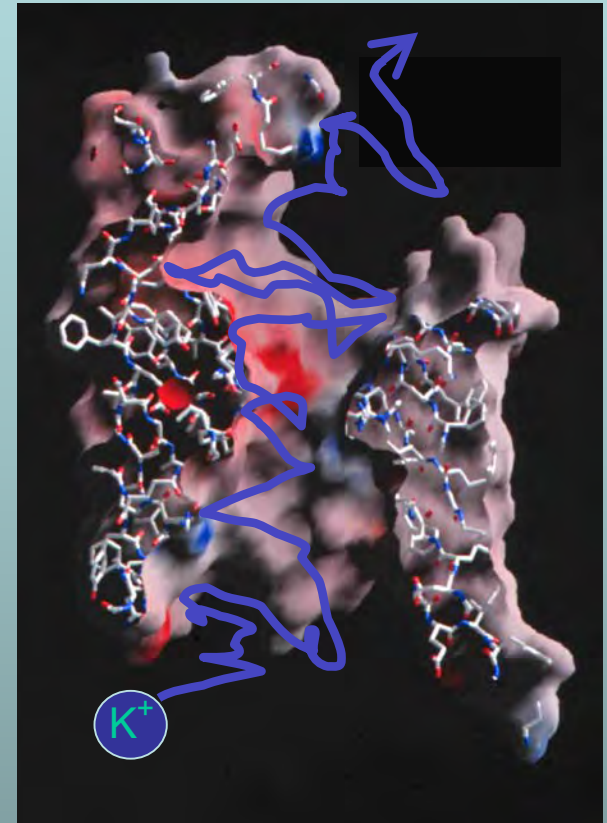
Are involved in secretion and absorption in all cells: kidney, intestine, liver, adrenal glands, etc.

Are involved in thousands of diseases and many drugs act on channels

Are proteins whose genes (blueprints) can be manipulated by molecular genetics

Have structures shown by x-ray crystallography in favorable cases

Can be described by mathematics in some cases

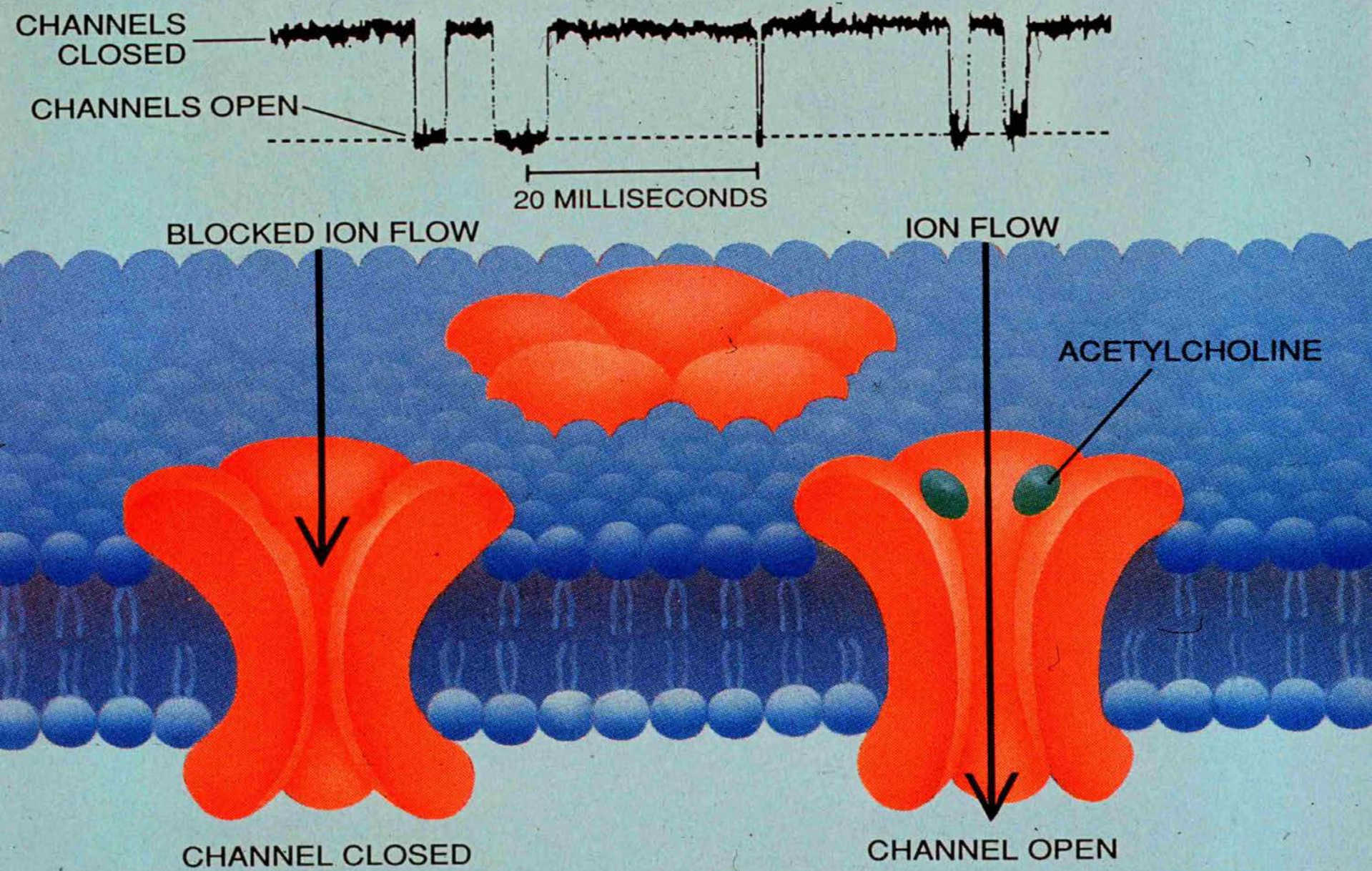


~30 x 10⁻⁹ meter

*Device is a Specific Word, that exploits specific mathematics & science

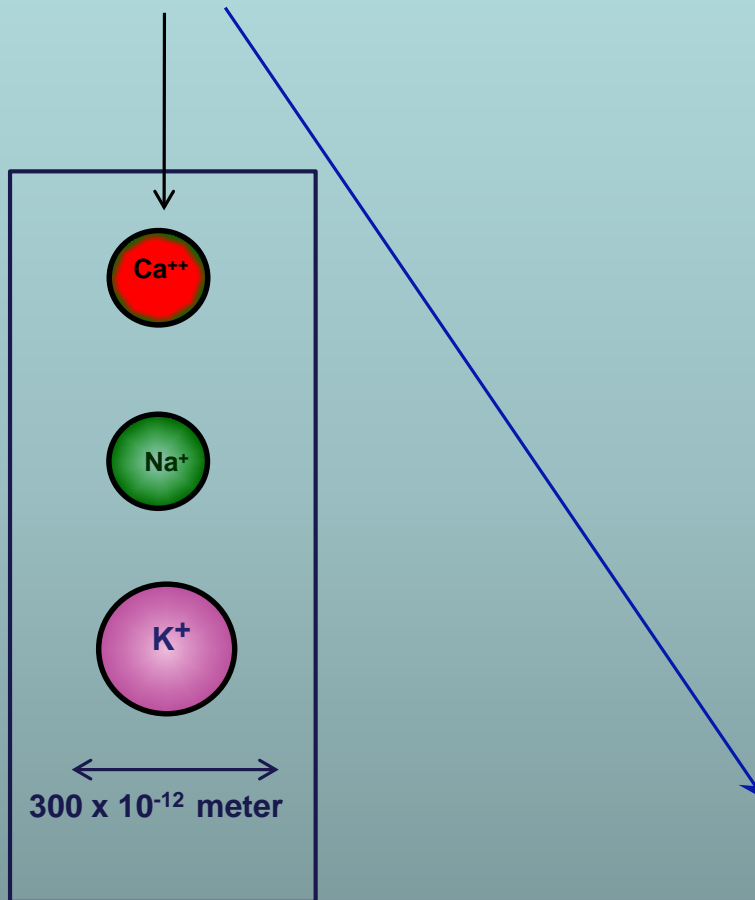
*nearly pico-valves: diameter is 400 – 900 x 10⁻¹² meter;
diameter of atom is ~200 x 10⁻¹² meter

Current in One Channel Molecule is a Random Telegraph Signal



Channels are Selective Molecular Devices

Different Ions Carry Different Signals through Different Channels

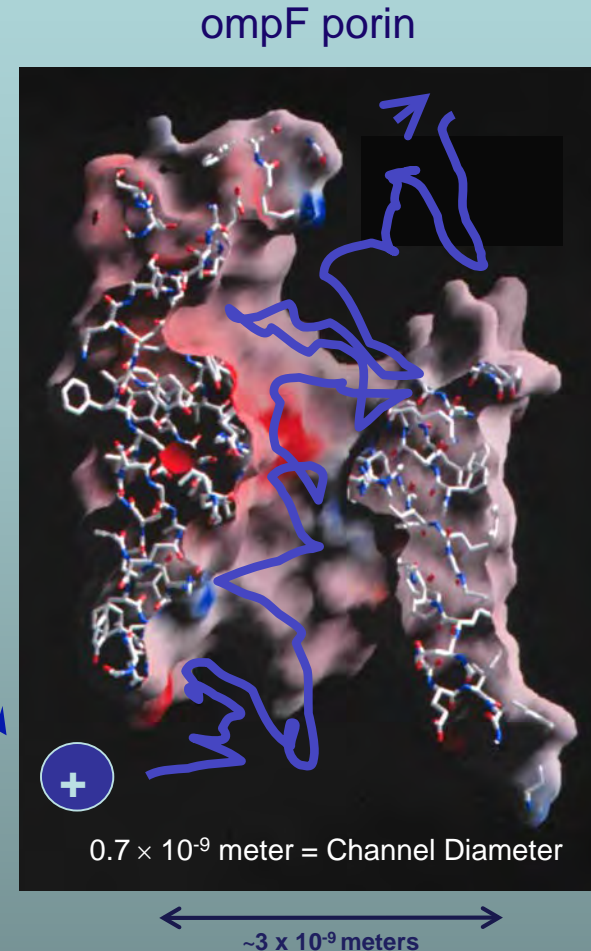


Diameter matters

Ionic solutions are NOT ideal

Classical Biochemistry assumes ideal solutions.

K^+ & Na^+ are identical only in Ideal Solutions.



Flow time scale is 10^{-4} sec to 1 min

Figure of ompF porin by Raimund Dutzler

*Engineering of Channels
by evolution makes them
Selective Devices*

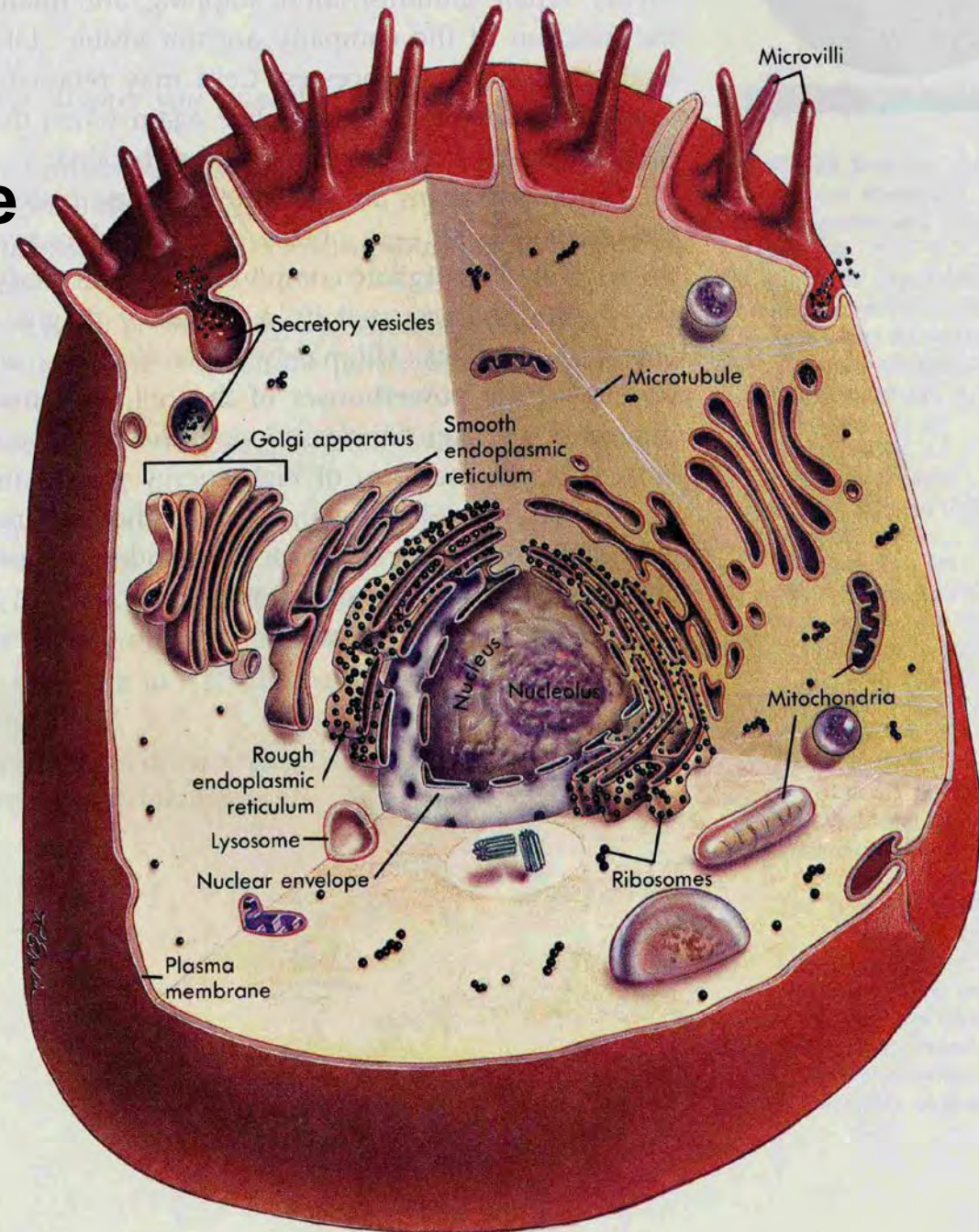
**Different Types of Channels
use
Different Types of Ions
for
Different Information**

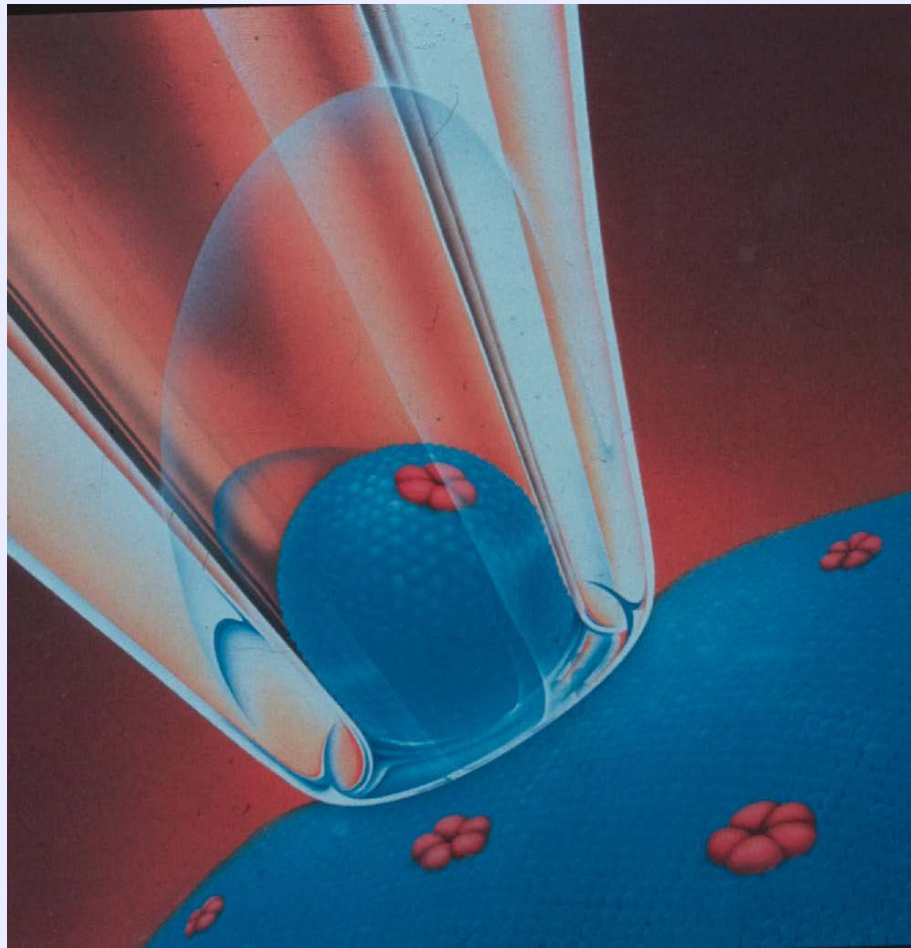
**All of life depends on the differences between
salts of potassium K^+ and sodium Na^+ .**

If cells cannot distinguish K^+ from Na^+ , they swell, burst and die.

The Cell

Note: intra-cellular compartments are defined by their membranes





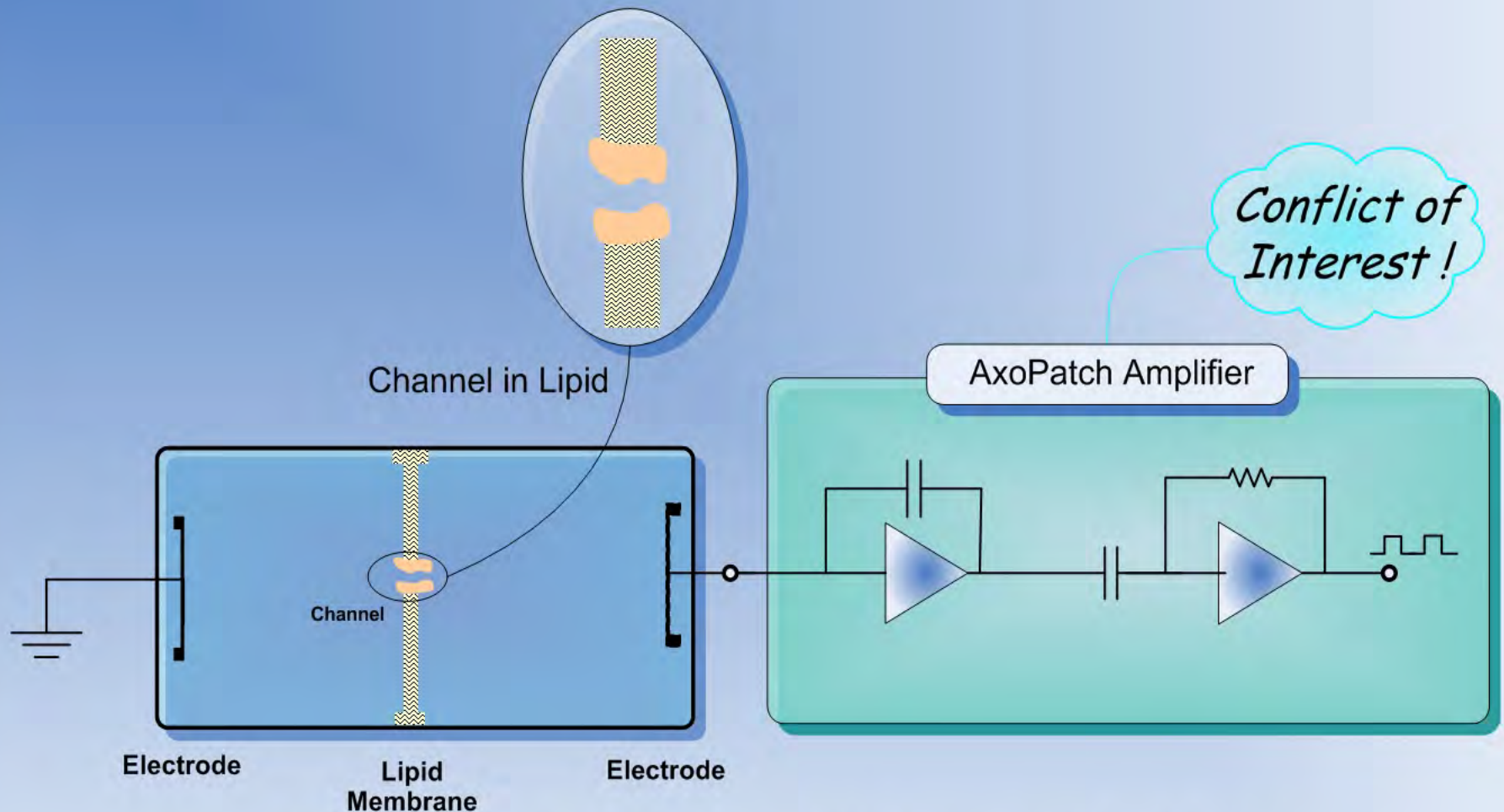
Patch clamp and **Bilayer apparatus** clamp **ion concentrations** in the baths and the **voltage** across membranes.



Patch Clamp Setup

Recordings from One Molecule

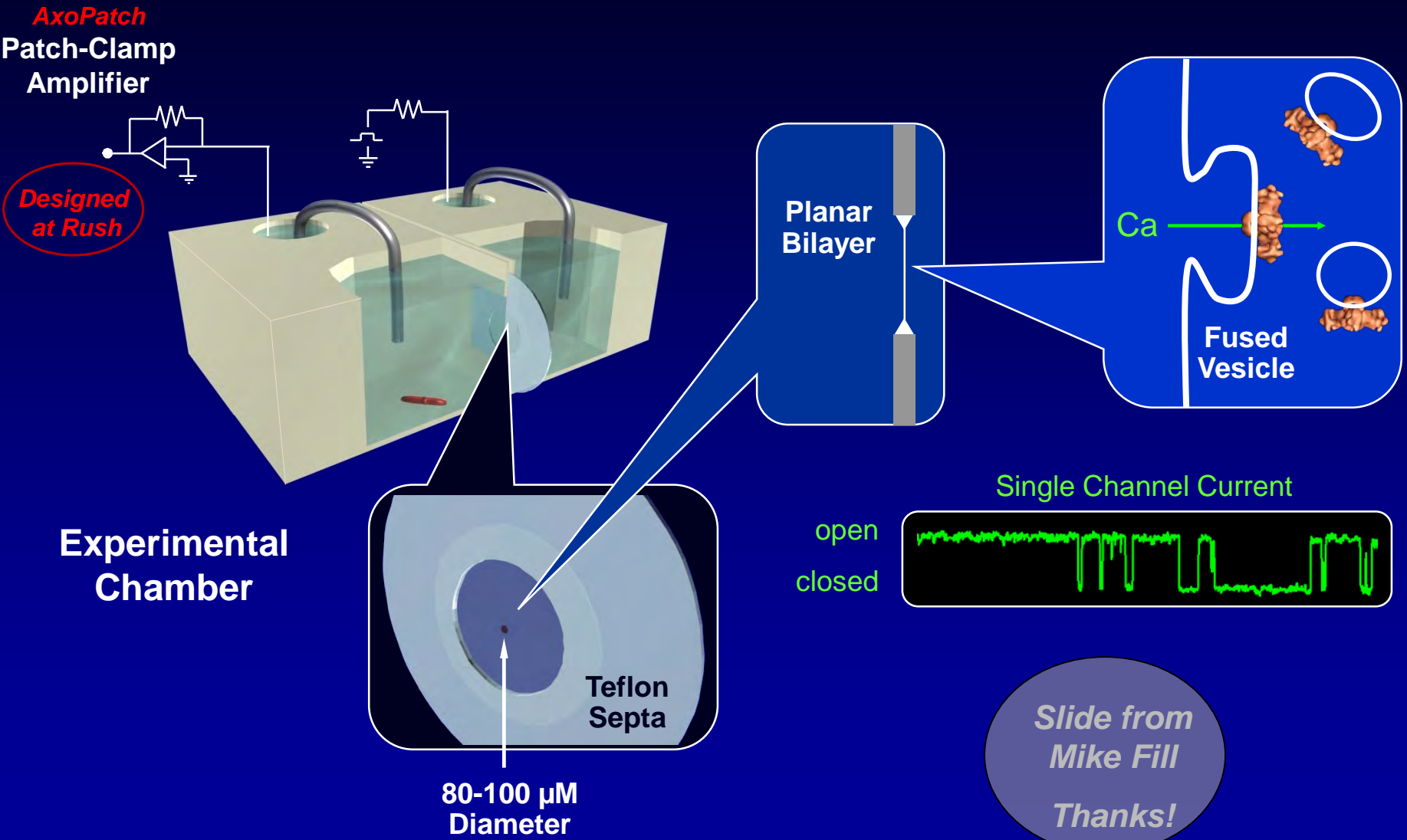
Single Channel Recording



Channel in Lipid Bilayer Set-up

Recording Current from One Protein Molecule

SINGLE isolated RyR Channels in Artificial Planar Lipid Bilayers



*Slide from
Mike Fill
Thanks!*

Thousands of Molecular Biologists Study Channels as Devices every day,

One protein molecule at a time

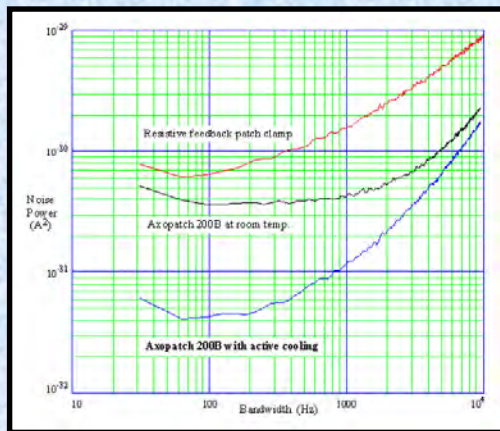
This number is not an exaggeration.

We have sold >10,000 AxoPatch amplifiers

AxoPatch 200B



Designed at Rush
Current Noise



Femto-amps
(10⁻¹⁵ A)

Ion Channel Monthly

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Popular publications for March ([view most recent](#))

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2. [The Angelman Syndrome Protein Ube3A Regulates Synapse Development by Ubiquitinating Arc](#). *Cell*
3. [AMPA receptors--another twist?](#) *Science*
4. [Molecular Basis of Calcium Signaling in Lymphocytes: STIM and ORAI](#). *Annu Rev Immunol*
5. [Neurological Channelopathies](#). *Annu Rev Neurosci*
6. [New antiarrhythmic drugs for treatment of atrial fibrillation](#). *Lancet*
7. [A Glial Signal Consisting of Gliomedin and NrCAM Clusters Axonal Na\(+\) Channels during the Formation of Nodes of Ranvier](#). *Neuron*
8. [Small Molecule Activators of TRPML3](#). *Chem Biol*
9. [Truncated \(beta\)-amyloid peptide channels provide an alternative mechanism for Alzheimer's Disease and Down syndrome](#). *Proc Natl Acad Sci U S A*
10. [Modelling the molecular mechanisms of synaptic plasticity using systems biology approaches](#). *Nat Rev Neurosci*

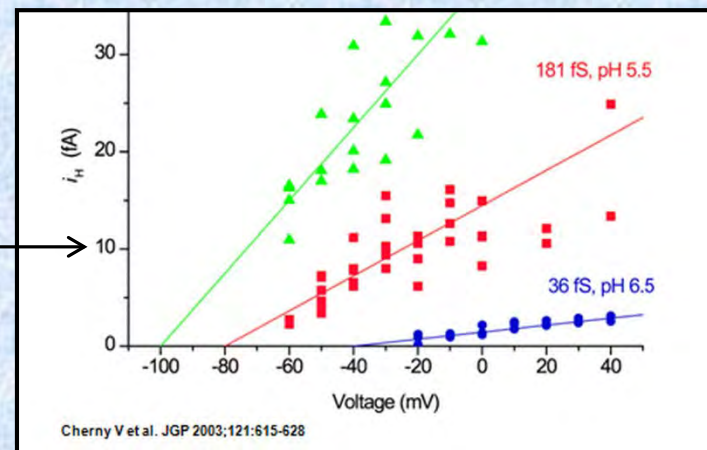
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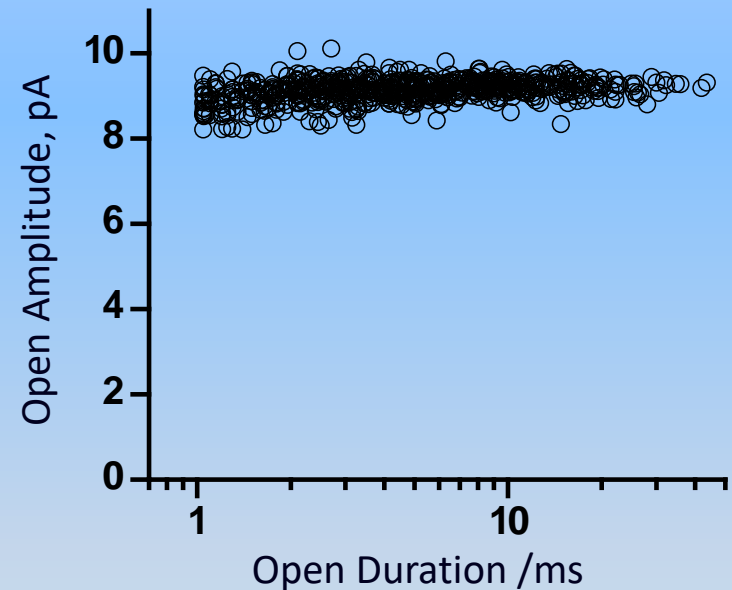
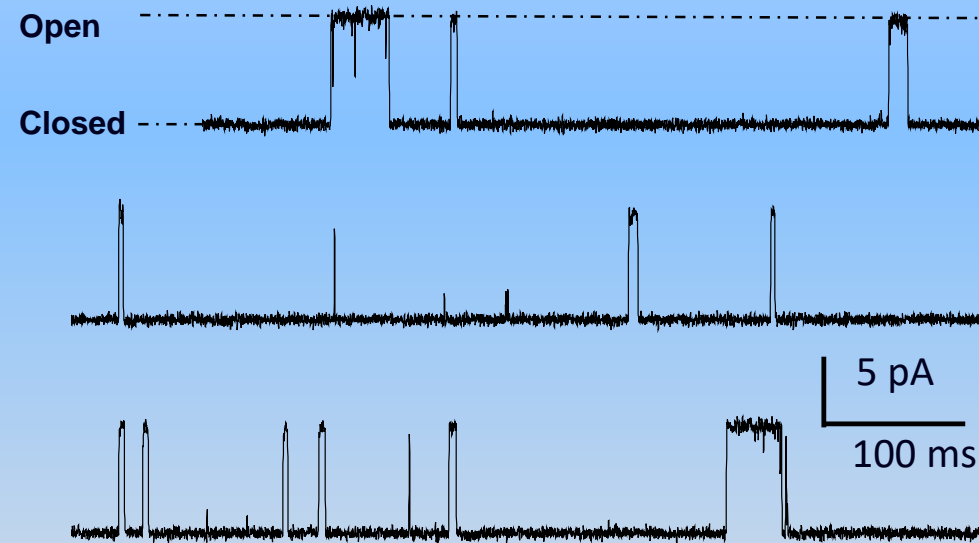
Current Noise



Channel Structure Does Not Change once the channel is open

Current vs. time

Amplitude vs. Duration



Lowpass Filter = 1 kHz Sample Rate = 20 kHz

Typical Raw Single Channel Records

Ca²⁺ Release Channel of Inositol Trisphosphate Receptor: slide and data from Josefina Ramos-Franco. Thanks!

Where to start?

Why not compute all the atoms?

Multi-Scale Issues

Journal of Physical Chemistry C (2010)114:20719

Computational Scale	Biological Scale	<u>Ratio</u>
<u>Time</u> 10^{-15} sec	10^{-4} sec	10^{11}
<u>Length</u> 10^{-11} m	10^{-5} m	10^6
<u>Spatial Resolution</u>	<i>Three Dimensional</i> $(10^4)^3$	10^{12}
<u>Volume</u> 10^{-30} m ³	$(10^{-4} \text{ m})^3 = 10^{-12} \text{ m}^3$	10^{18}
<u>Solute Concentration</u> <i>including Ca²⁺ mixtures</i>	10^{-11} to 10^1 M	10^{12}

Atomic and Macro Scales are BOTH used by channels because they are nanovalves
so atomic and macro scales must be
Computed and CALIBRATED Together

This may be impossible in all-atom simulations

**Multi-Scale Issues
are
Always Present
in
Atomic Scale Engineering**

**Atomic & Macro Scales are
both used by channels
just because
Channels are Nanovalves**

By definition: all valves use small structures to control large flows

Atomic Scale Engineering

uses a

Few Atoms

to

Control

Macroscopic Flows

so **Atomic** and **Macro Scales** must be

Computed

and

CALIBRATED

together

All Life Occurs in Ionic Mixtures

in which $[\text{Ca}^{2+}]$ is important* as a control signal

**Simulations must deal with
Multiple Components**

as well as
Multiple Scales

This may be nearly impossible for ionic mixtures

because

‘everything’ interacts with ‘everything else’

on both atomic and macroscopic scales

particularly when mixtures flow

* $[\text{Ca}^{2+}]$ ranges from 1×10^{-8} M inside cells to 10 M inside channels

**Uncalibrated Simulations
will make devices
that do not work**

Details matter in devices

Where to start?

Mathematically ?

Physically ?

Reduced Models are Needed

Reduced Models are Device Equations
like Input Output Relations of Engineering Systems

The device equation is the mathematical
statement of how the system works

Device Equations describe 'Slow Variables'
found in some complicated systems

How find a Reduced Model?

Biology is Easier than Physics

Reduced Models Exist*

for important biological functions
or the

**Animal would not survive
to reproduce**

*Evolution provides the existence theorems and uniqueness conditions
so hard to find in theory of inverse problems.

*(Some biological systems – the human shoulder – are not robust,
probably because they are incompletely evolved,
i.e. they are in a local minimum 'in fitness landscape'.*

I do not know how to analyze these.

I can only describe them in the classical biological tradition.)

Multi-scale Engineering
is
MUCH easier when robust
Reduced Models Exist

Reduced models exist
because
they are the adaptation
created by evolution
to perform a biological function
like selectivity

Reduced Models
and its parameters
are found by
Inverse Methods
of Reverse Engineering

Bioengineers: this is reverse engineering

Inverse Problems

Given the Output Determine the Reduced Model

For example,

**Find Charge Distribution
in Channel
from
Current Voltage Relations**

Problem (with noise and systematic error) has
actually been solved by Tikhonov Regularization

Burger, Eisenberg, Engl (2007) SIAM J Applied Math 67: 960-989

using procedures developed by Engl to study Blast Furnaces and their Explosions

Bioengineers: this is reverse engineering

Inverse Problems

Find the Model, given the Output

*Many answers are possible: 'ill posed' **

Central Issue

Which answer is right?

*Ill posed problems with too little data seem complex, even if they are not. Some of biology seems complex for that reason. The question is which 'some'?

Inverse Problems: many answers possible

Central Issue

Which answer is right?

Key is

ALWAYS

Large Amount of Data

from

Many Different Conditions

*Almost too much data was available for reduced model:
Burger, Eisenberg and Engl (2007) SIAM J Applied Math 67:960-989*

Inverse Problems: many answers possible

Which answer is right?

Key is

Large Amount of Data from **Many Different Conditions**

Otherwise problem is 'ill-posed' and has no answer or even set of answers

Molecular Dynamics

usually yields

ONE data point

at one concentration

MD is not yet well calibrated

(i.e., for activity = free energy per mole)

for Ca²⁺ or ionic mixtures like seawater or biological solutions

Working Hypothesis:

Crucial Biological Adaptation is
Crowded Ions *and* Side Chains

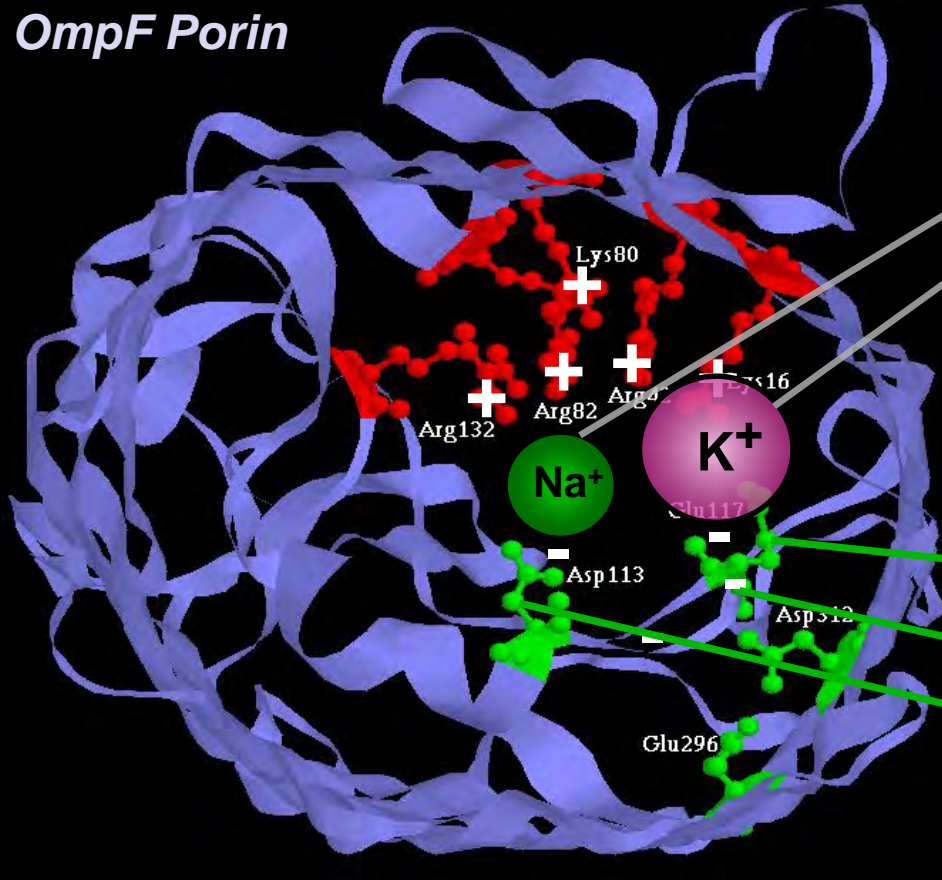
**Wise to use the Biological Adaptation
to make the reduced model!**

Reduced Models allow much easier Atomic Scale Engineering

Active Sites of Proteins are Very Charged

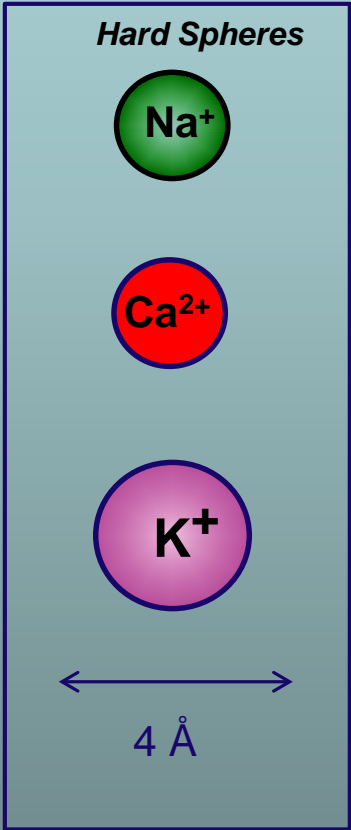
7 charges ~ 20 M net charge = $1.2 \times 10^{22} \text{ cm}^{-3}$

liquid **Water** is **55 M**
solid **NaCl** is **37 M**



Ions are Crowded

Induced Fit of Side Chains



Selectivity Filters and Gates of Ion Channels are **Active Sites**

Figure adapted from Tilman Schirmer

Crowded Active Sites

in 573 Enzymes

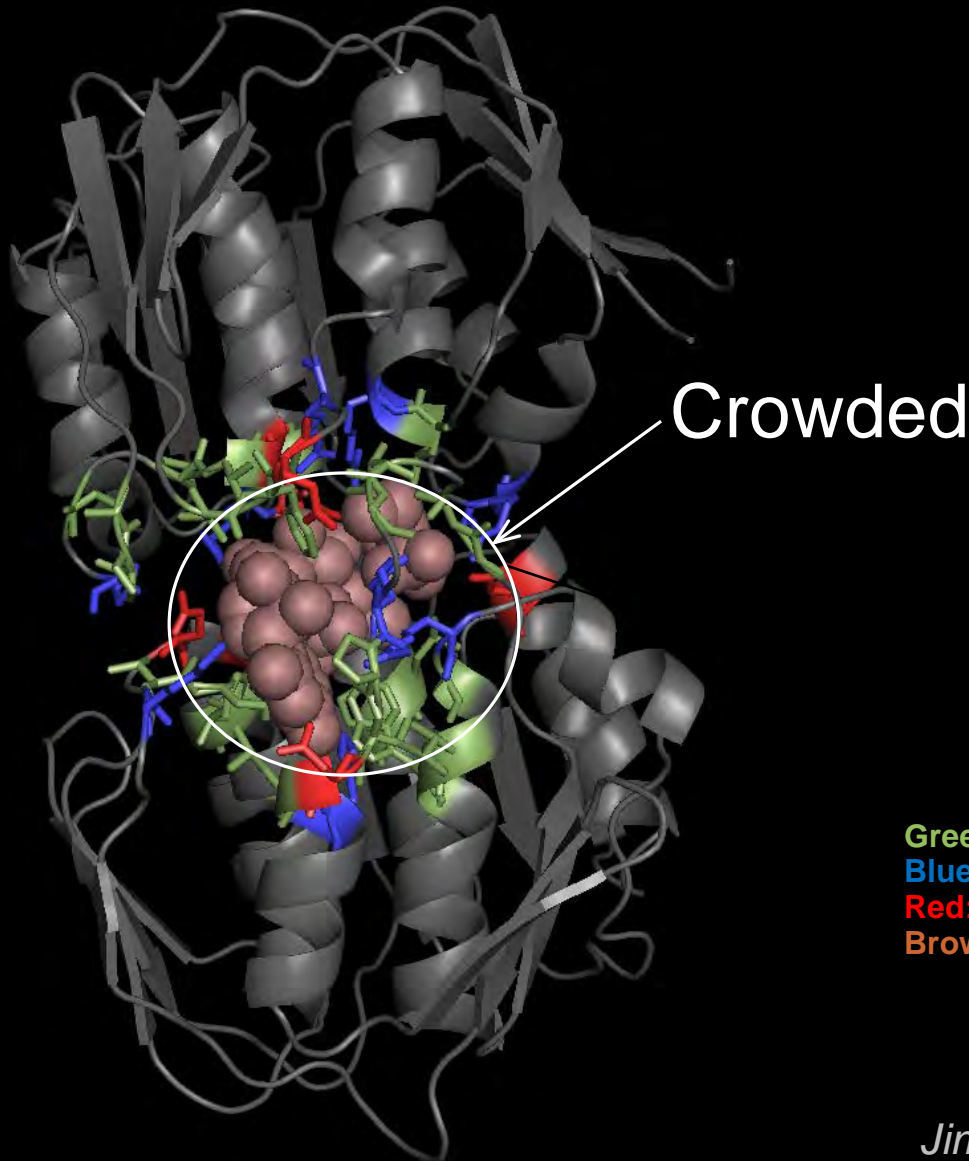
Enzyme Type		Catalytic Active Site			Protein
		Density (Molar)			Elsewhere
		<i>Acid</i> (positive)	<i>Basic</i> (negative)	 Total 	
	Total (n = 573)	10.6	8.3	18.9	2.8
EC1	Oxidoreductases (n = 98)	7.5	4.6	12.1	2.8
EC2	Transferases (n = 126)	9.5	7.2	16.6	3.1
EC3	Hydrolases (n = 214)	12.1	10.7	22.8	2.7
EC4	Lyases (n = 72)	11.2	7.3	18.5	2.8
EC5	Isomerases (n = 43)	12.6	9.5	22.1	2.9
EC6	Ligases (n = 20)	9.7	8.3	18.0	3.0

EC2: TRANSFERASES

Average Ionizable Density: 19.8 Molar

Example
UDP-N-ACETYLGLUCOSAMINE
ENOLPYRUVYL TRANSFERASE
(PDB:1UAE)

Functional Pocket Volume: 1462.40 Å³
Density : 19.3 Molar (11.3 M+ . 8 M-)



Green: Functional pocket residues
Blue: Basic = Probably Positive = R+K+H
Red: Acid = Probably Negative = E + Q
Brown Uridine-Diphosphate-N-acetylglucosamine

Everything Interacts with **Everything Else**

by steric exclusion
inside crowded active sites

**Everything interacts with macroscopic Boundary Conditions
(and much else)
through long range electric field**

'Law' of mass action needs to be generalized

Three Channel Types

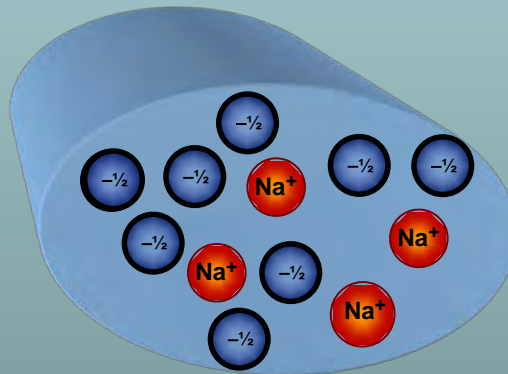
RyR, $\text{Ca}_v = \text{EEEE}$, and $\text{Na}_v = \text{DEKA}$

analyzed successfully*

in a wide range of solutions by the

'All Spheres' Primitive Model

Implicit solvent model of open channel



ions and protein side chains are hard spheres
in this model

* Many methods have been used in more than 30 papers
since Nonner and Eisenberg, 1998

Best Evidence is from the
RyR Receptor

Dirk Gillespie

Dirk_Gillespie@rush.edu



Gerhard Meissner, Le Xu, et al,
not Bob Eisenberg

- **More than 120 combinations of solutions & mutants**
- **7 mutants with significant effects fit successfully**

1. Gillespie, D., Energetics of divalent selectivity in a calcium channel: the ryanodine receptor case study. *Biophys J*, 2008. 94(4): p. 1169-1184.
2. Gillespie, D. and D. Boda, Anomalous Mole Fraction Effect in Calcium Channels: A Measure of Preferential Selectivity. *Biophys. J.*, 2008. 95(6): p. 2658-2672.
3. Gillespie, D. and M. Fill, Intracellular Calcium Release Channels Mediate Their Own Countercurrent: Ryanodine Receptor. *Biophys. J.*, 2008. 95(8): p. 3706-3714.
4. Gillespie, D., W. Nonner, and R.S. Eisenberg, Coupling Poisson-Nernst-Planck and Density Functional Theory to Calculate Ion Flux. *Journal of Physics (Condensed Matter)*, 2002. 14: p. 12129-12145.
5. Gillespie, D., W. Nonner, and R.S. Eisenberg, Density functional theory of charged, hard-sphere fluids. *Physical Review E*, 2003. 68: p. 0313503.
6. Gillespie, D., Valisko, and Boda, Density functional theory of electrical double layer: the RFD functional. *Journal of Physics: Condensed Matter*, 2005. 17: p. 6609-6626.
7. Gillespie, D., J. Giri, and M. Fill, Reinterpreting the Anomalous Mole Fraction Effect. The ryanodine receptor case study. *Biophysical Journal*, 2009. 97: p. pp. 2212 - 2221
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9. Gillespie, D., D. Boda, Y. He, P. Apel, and Z.S. Siwy, Synthetic Nanopores as a Test Case for Ion Channel Theories: The Anomalous Mole Fraction Effect without Single Filing. *Biophys. J.*, 2008. 95(2): p. 609-619.
10. Malasics, A., D. Boda, M. Valisko, D. Henderson, and D. Gillespie, Simulations of calcium channel block by trivalent cations: Gd(3+) competes with permeant ions for the selectivity filter. *Biochim Biophys Acta*, 2010. 1798(11): p. 2013-2021.
11. Roth, R. and D. Gillespie, Physics of Size Selectivity. *Physical Review Letters*, 2005. 95: p. 247801.
12. Valisko, M., D. Boda, and D. Gillespie, Selective Adsorption of Ions with Different Diameter and Valence at Highly Charged Interfaces. *Journal of Physical Chemistry C*, 2007. 111: p. 15575-15585.
13. Wang, Y., L. Xu, D. Pasek, D. Gillespie, and G. Meissner, Probing the Role of Negatively Charged Amino Acid Residues in Ion Permeation of Skeletal Muscle Ryanodine Receptor. *Biophysical Journal*, 2005. 89: p. 256-265.
14. Xu, L., Y. Wang, D. Gillespie, and G. Meissner, Two Rings of Negative Charges in the Cytosolic Vestibule of T Ryanodine Receptor Modulate Ion Fluxes. *Biophysical Journal*, 2006. 90: p. 443-453.

Solved by DFT-PNP (Poisson Nernst Planck)

DFT-PNP
gives location
of Ions and 'Side Chains'
as OUTPUT

Other methods

give nearly identical results

DFT (Density Functional Theory of fluids, *not electrons*)

MMC (Metropolis Monte Carlo)

SPM (Primitive Solvent Model)

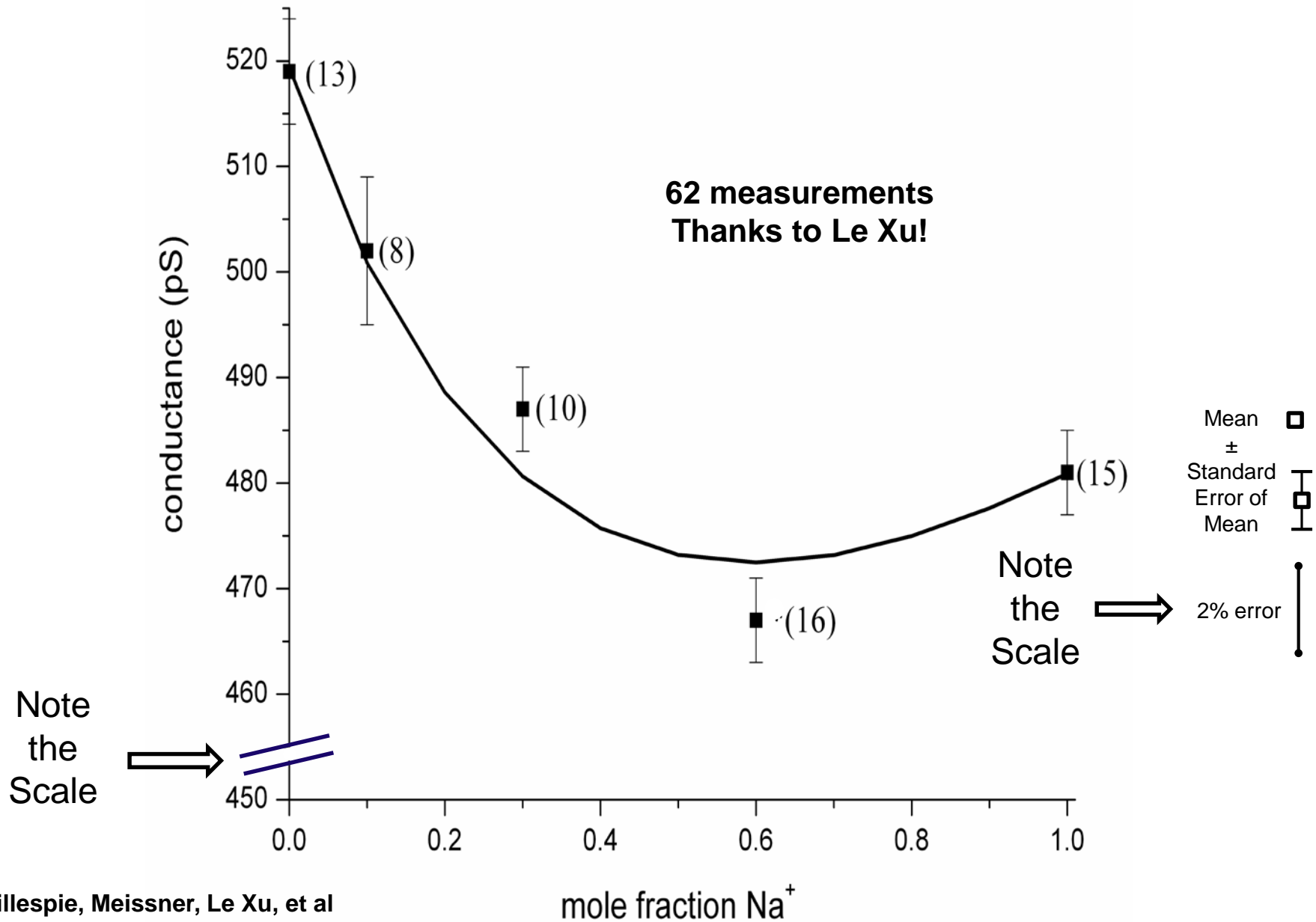
EnVarA (Energy Variational Approach)

Non-equil MMC (Boda, Gillespie) several forms

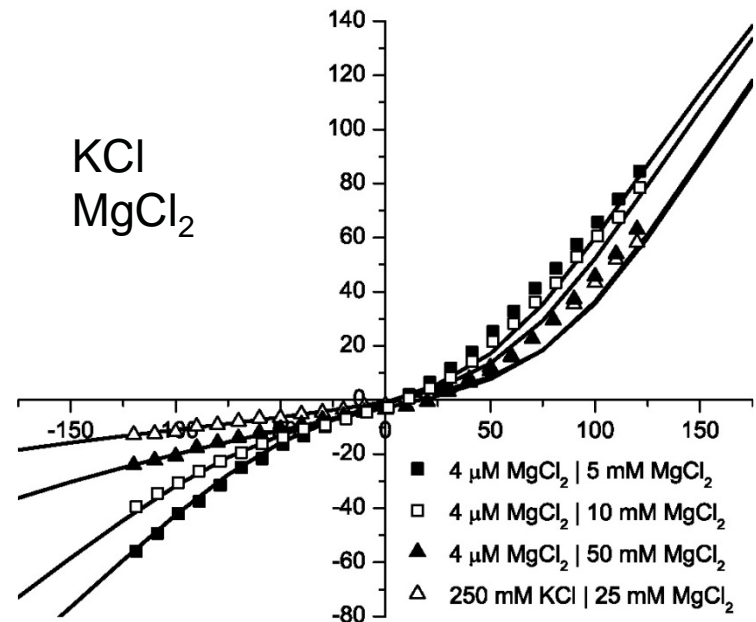
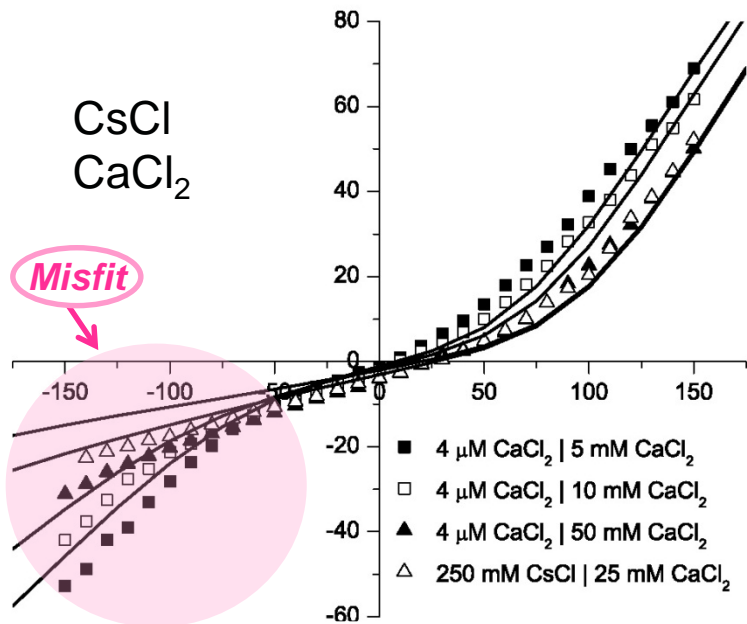
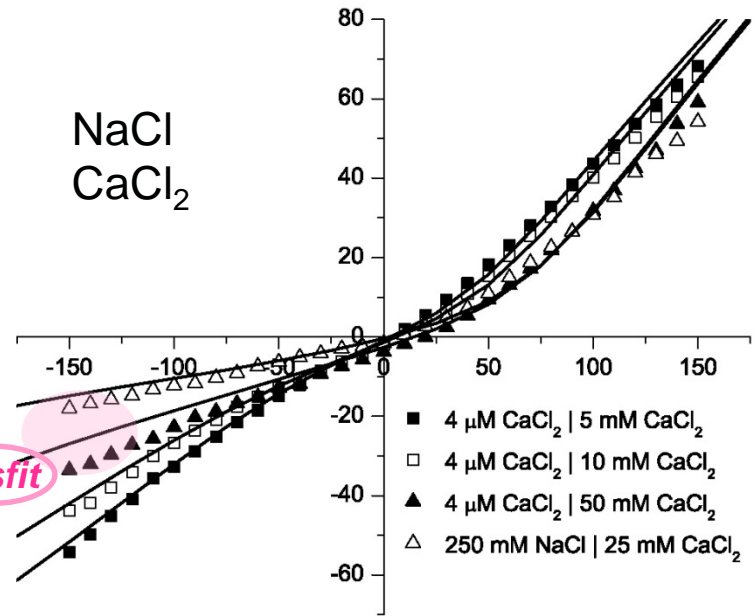
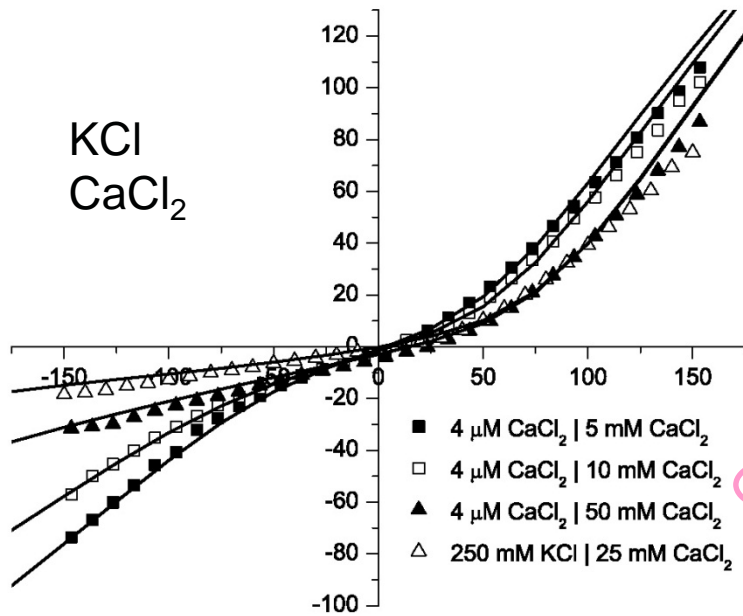
Steric PNP (simplified EnVarA)

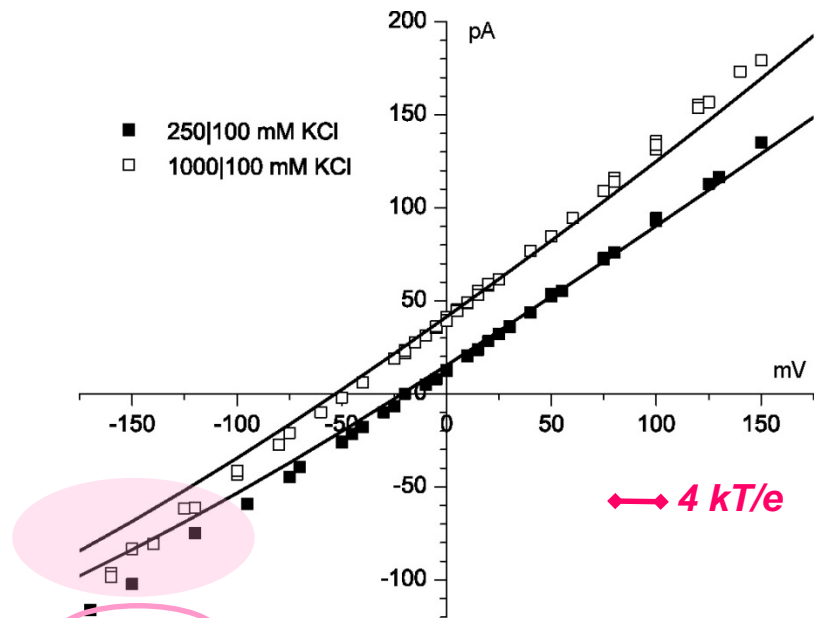
Poisson Fermi

The model predicted an AMFE for Na⁺/Cs⁺ mixtures before it had been measured

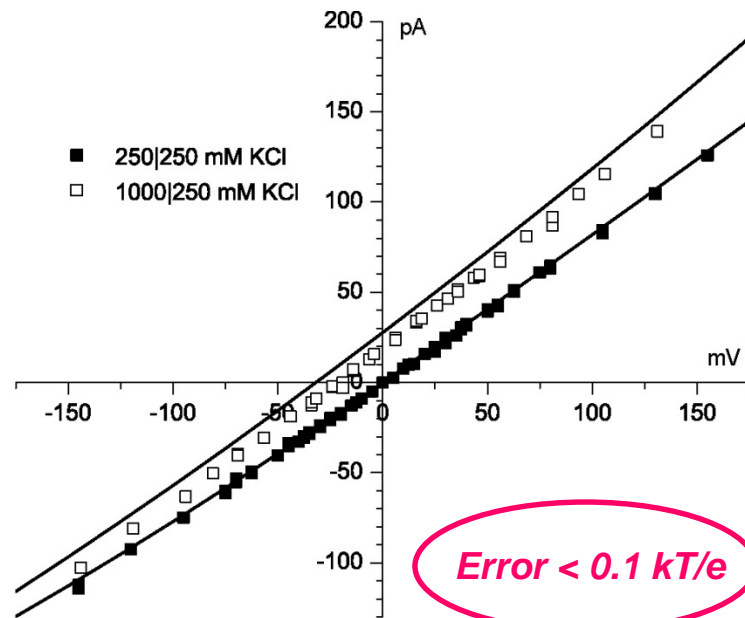


Divalents

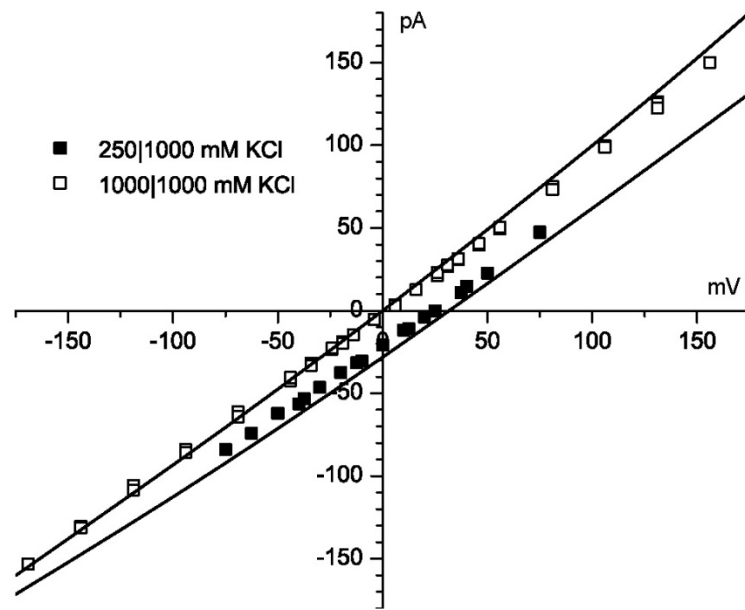
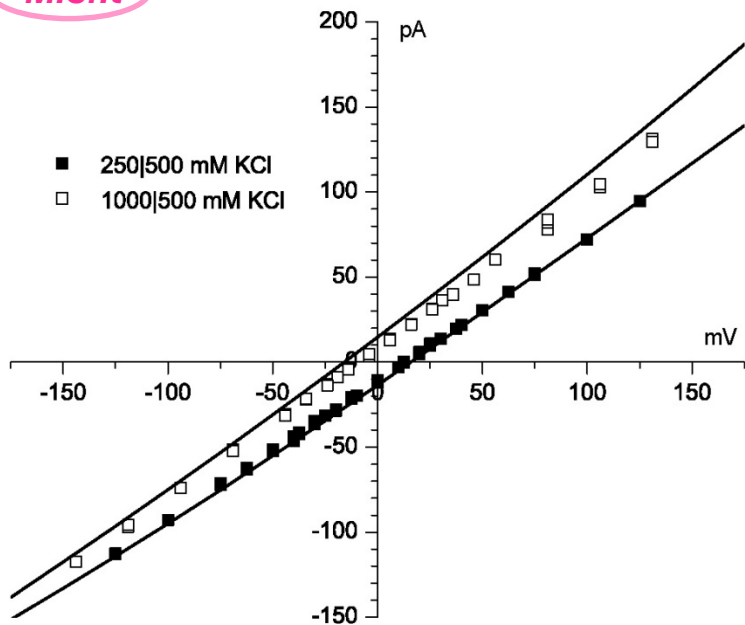




Misfit

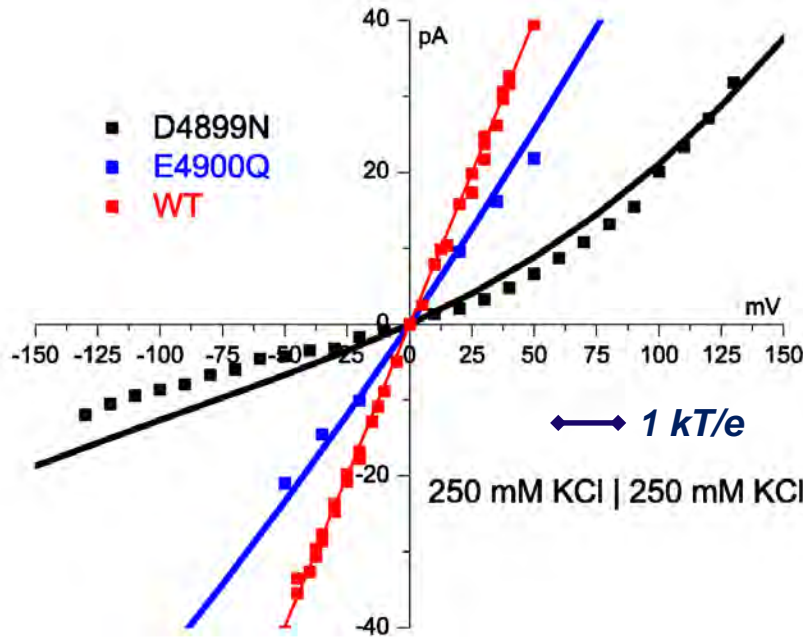


Error < 0.1 kT/e

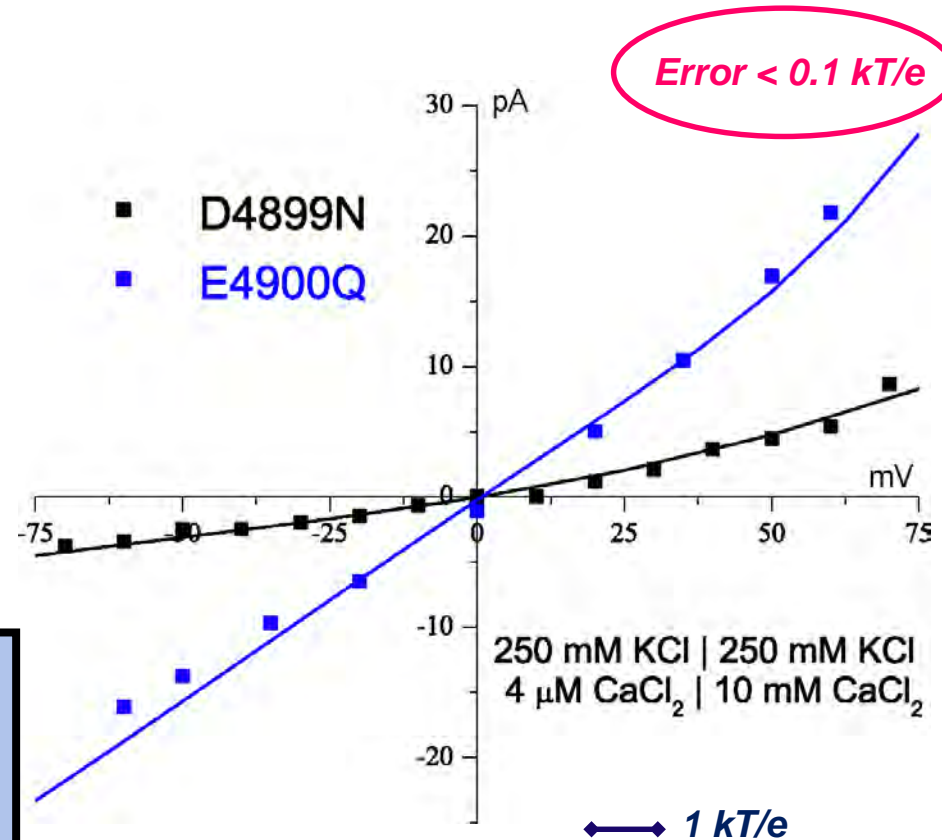


Theory fits Mutation with Zero Charge

Theory Fits Mutant in K



Theory Fits Mutant in K + Ca



Protein charge density

wild type* **13 M** ⇒ **0 M** in D4899

Solid Na⁺Cl⁻ is 37 M

*some wild type curves not shown, 'off the graph'

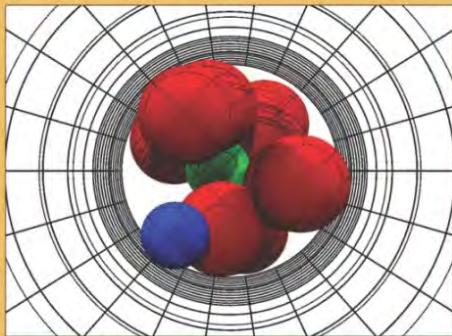
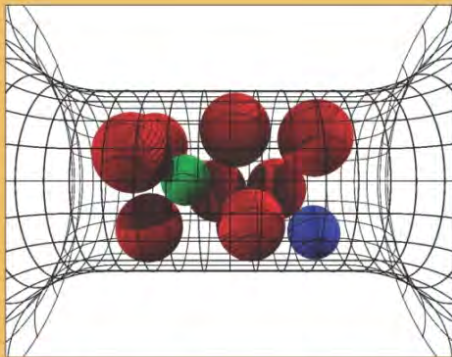
Gillespie *et al*

J Phys Chem 109 15598 (2005)

Calcium Channel

JGP

The Journal of General Physiology
Vol 133 • No 5 • May 2009



www.jgp.org

- Nonner, W., D. P. Chen, and B. Eisenberg. 1998. Anomalous Mole Fraction Effect, Electrostatics, and Binding in Ionic Channels. *Biophysical Journal* 74:2327-2334.
- Nonner, W., L. Catacuzzeno, and B. Eisenberg. 2000. Binding and Selectivity in L-type Ca Channels: a Mean Spherical Approximation. *Biophysical Journal* 79:1976-1992.
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- Boda, D., W. Nonner, D. Henderson, B. Eisenberg, and D. Gillespie. 2008. Volume exclusion in calcium selective channels. *Biophys. J.* biophysj.107.122796.
- Boda, D., M. Valisko, B. Eisenberg, W. Nonner, D. Henderson, and D. Gillespie. 2006. Effect of Protein Dielectric Coefficient on the Ionic Selectivity of a Calcium Channel. *Journal of Chemical Physics* 125:034901.
- Boda, D., T. Varga, D. Henderson, D. Busath, W. Nonner, D. Gillespie, and B. Eisenberg. 2004. Monte Carlo simulation study of a system with a dielectric boundary: application to calcium channel selectivity. *Molecular Simulation* 30:89-96.
- Boda, D., M. Valisko, B. Eisenberg, W. Nonner, D. Henderson, and D. Gillespie. 2007. The combined effect of pore radius and protein dielectric coefficient on the selectivity of a calcium channel. *Physical Review Letters* 98:168102.

More than 35 papers are available at

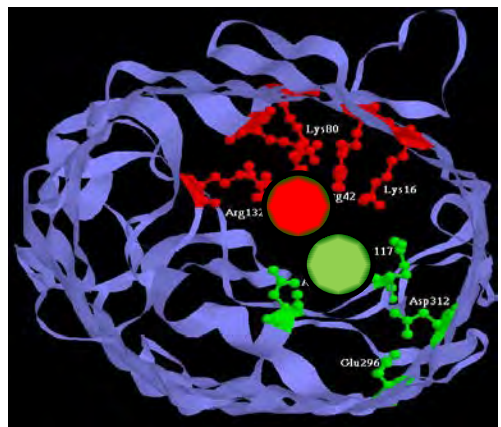
ftp://ftp.rush.edu/users/molebio/Bob_Eisenberg/reprints

<http://www.phys.rush.edu/RSEisenberg/physioeis.html>

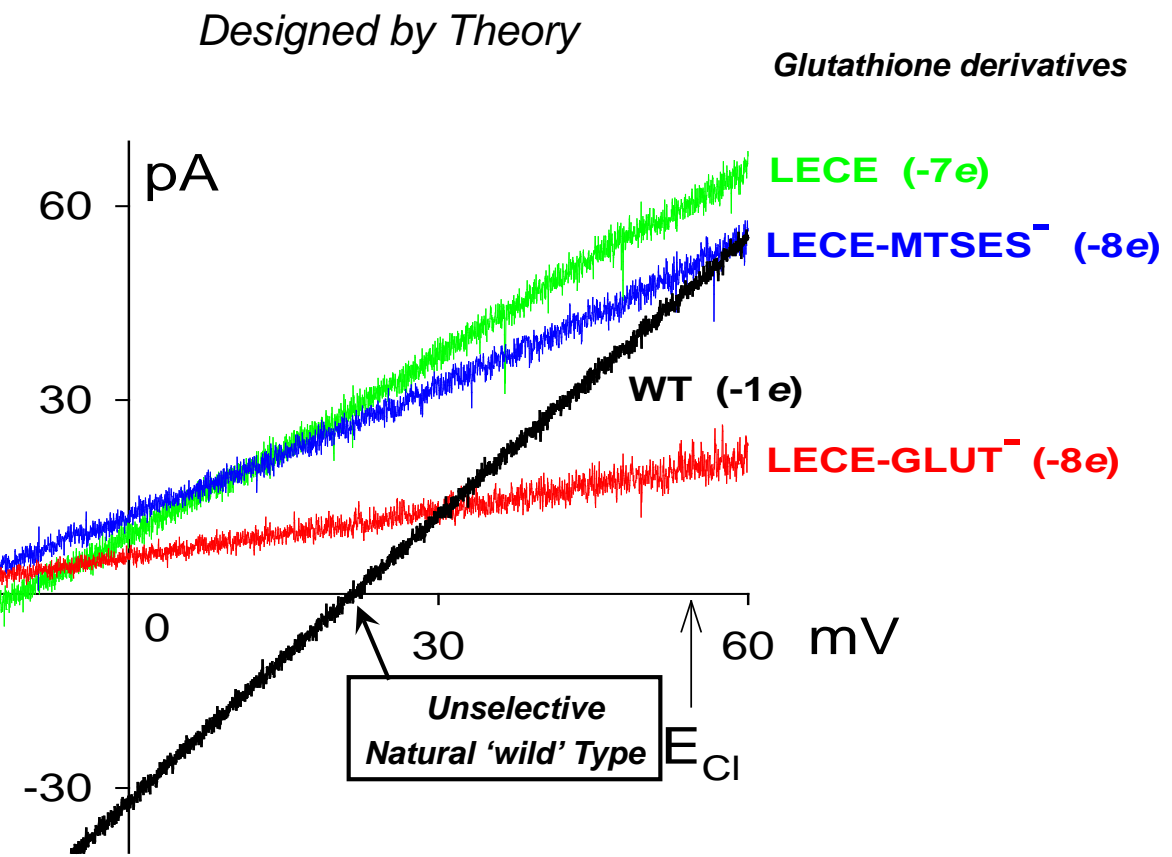
Experiments have 'engineered' channels (5 papers) including

Two Synthetic Calcium Channels

Atomic Scale



Calcium selective



As density of permanent charge increases, channel becomes calcium selective

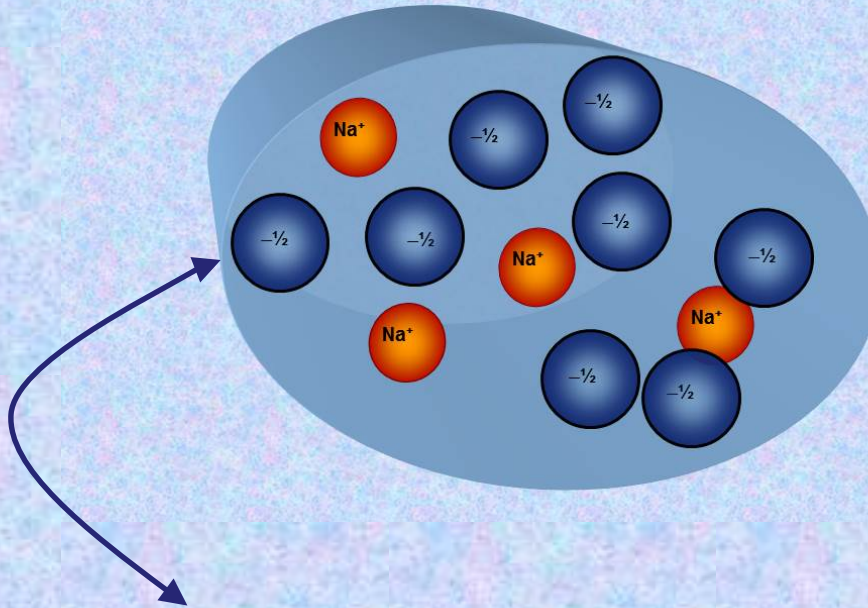
$E_{rev} \rightarrow E_{Ca}$ in 0.1M || 1.0 M CaCl₂; pH 8.0

Macro Scale

built by Henk Miedema, Wim Meijberg of BioMade Corp. Groningen, Netherlands 45

Miedema et al, Biophys J 87: 3137-3147 (2004); 90:1202-1211 (2006); 91:4392-4400 (2006)

'All Spheres' Model



Side Chains are Spheres

Channel is a Cylinder

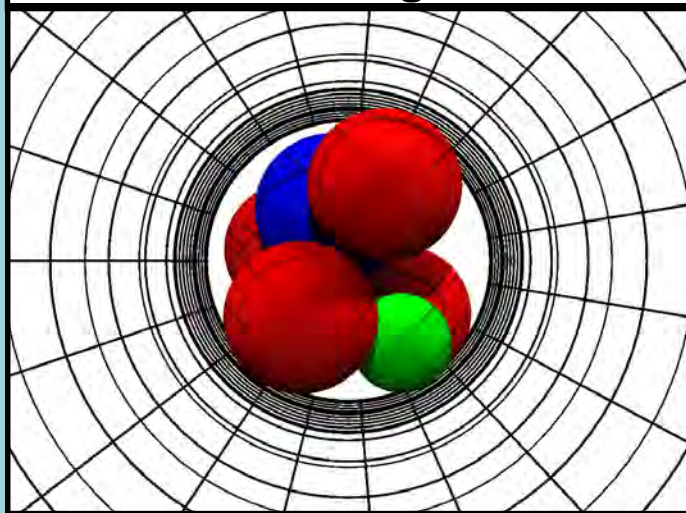
Side Chains are free to move within Cylinder

Ions and Side Chains are at free energy minimum

i.e., ions and side chains are 'self organized',

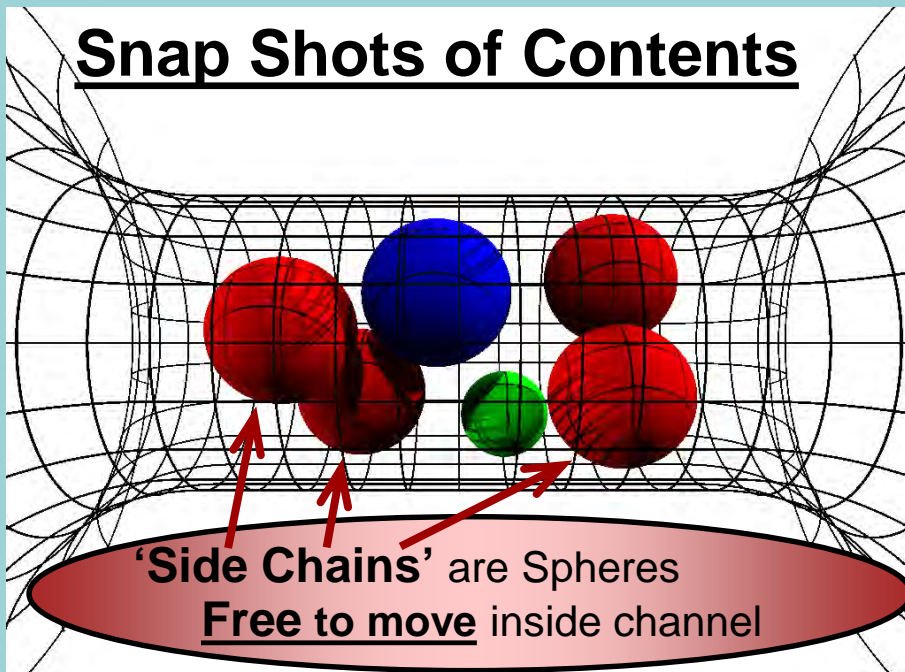
'Binding Site' is induced by substrate ions

Radial Crowding is Severe



6 Å

Snap Shots of Contents



Crowded Ions

Ion Diameters

'Pauling' Diameters

Ca ⁺⁺	1.98 Å
Na ⁺	2.00 Å
K ⁺	2.66 Å

'Side Chain' Diameter

Lysine K	3.00 Å
D or E	2.80 Å

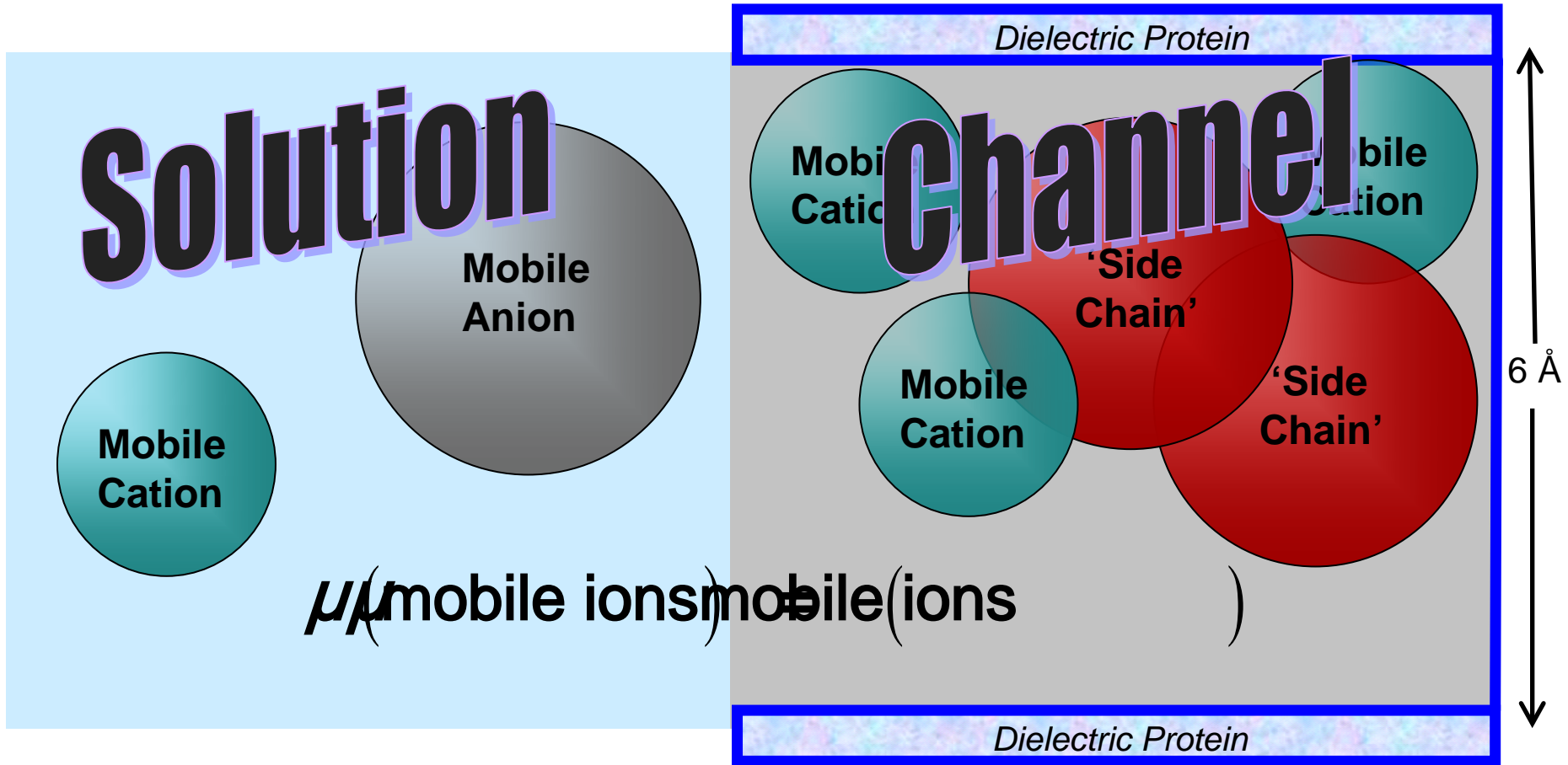
Channel Diameter 6 Å

Parameters are Fixed in all calculations
in all solutions for all mutants

Experiments and Calculations done at pH 8

47

Ion 'Binding' in Crowded Channel



Classical Donnan Equilibrium of Ion Exchanger

large mechanical forces

Side chains move within channel to their equilibrium position of minimal free energy.

We compute the Tertiary Structure as the structure of minimal free energy.

Metropolis Monte Carlo Simulates Location of Ions

both the mean and the variance

MMC *details*

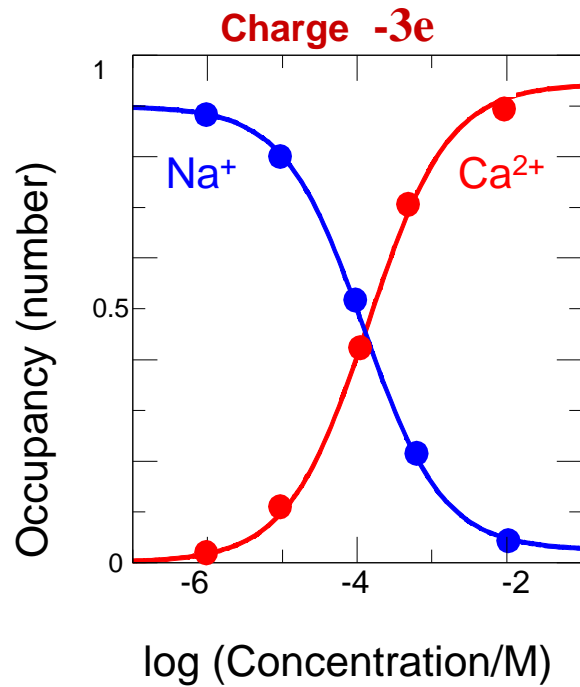
- 1) Start with Configuration A , with computed energy E_A
- 2) Move an ion to location B , with computed energy E_B
- 3) If spheres overlap, $E_B \rightarrow \infty$ and configuration is rejected
- 4) If spheres do not overlap, $E_B \neq 0$ and configuration may be accepted
 - (4.1) If $E_B < E_A$: accept new configuration.
 - (4.2) If $E_B > E_A$: accept new configuration with probability $\exp\left[-(E_A - E_B)/k_B T\right]$

Key idea

Instead of choosing configurations from uniform distribution, then weighting them with $\exp(-E/k_B T)$, **MMC** chooses them with a Boltzmann probability and weights them evenly.

Ca Channel

E
E
E
A



EEEE has full biological selectivity
in similar simulations

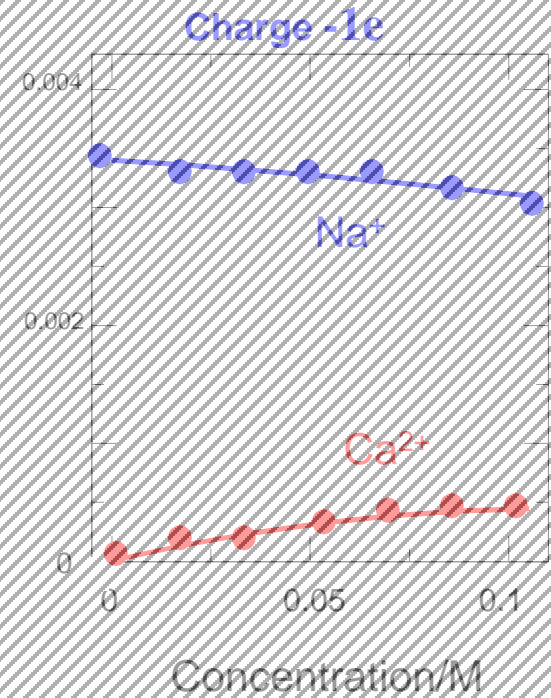
Mutation



Same Parameters

Na Channel

D
E
K
A



Boda, et al

Selectivity

comes from

Electrostatic Interaction

and

Steric Competition for Space



Repulsion

Location and Strength of Binding Sites
Depend on Ionic Concentration and
Temperature, etc

Rate Constants are Variables

Sodium Channel

Voltage controlled channel responsible for signaling in nerve and coordination of muscle contraction

Challenge

from channologists

Walter Stühmer and **Stefan Heinemann**

Göttingen

Leipzig

Max Planck Institutes

**Can THEORY explain the MUTATION
Calcium Channel into Sodium Channel?**

DEEA  **DEKA**

*Calcium
Channel*

*Sodium
Channel*

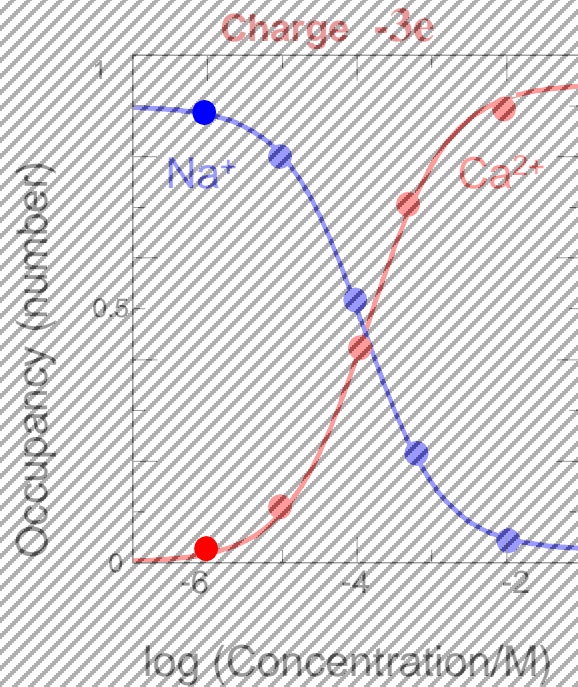
Ca Channel

Mutation
Same Parameters

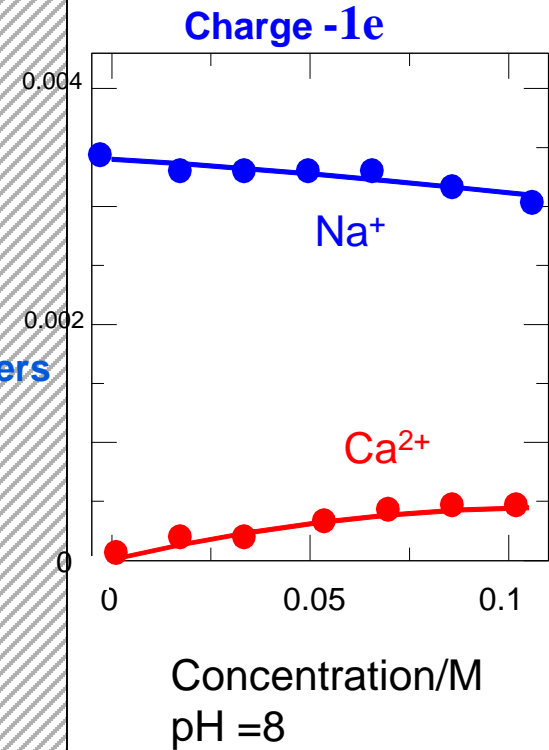
Na Channel

E
E
E
A

D
E
K
A



Mutation
Same Parameters
pH 8



EEEE has full biological selectivity
in similar simulations

Monte Carlo simulations of Boda, et al

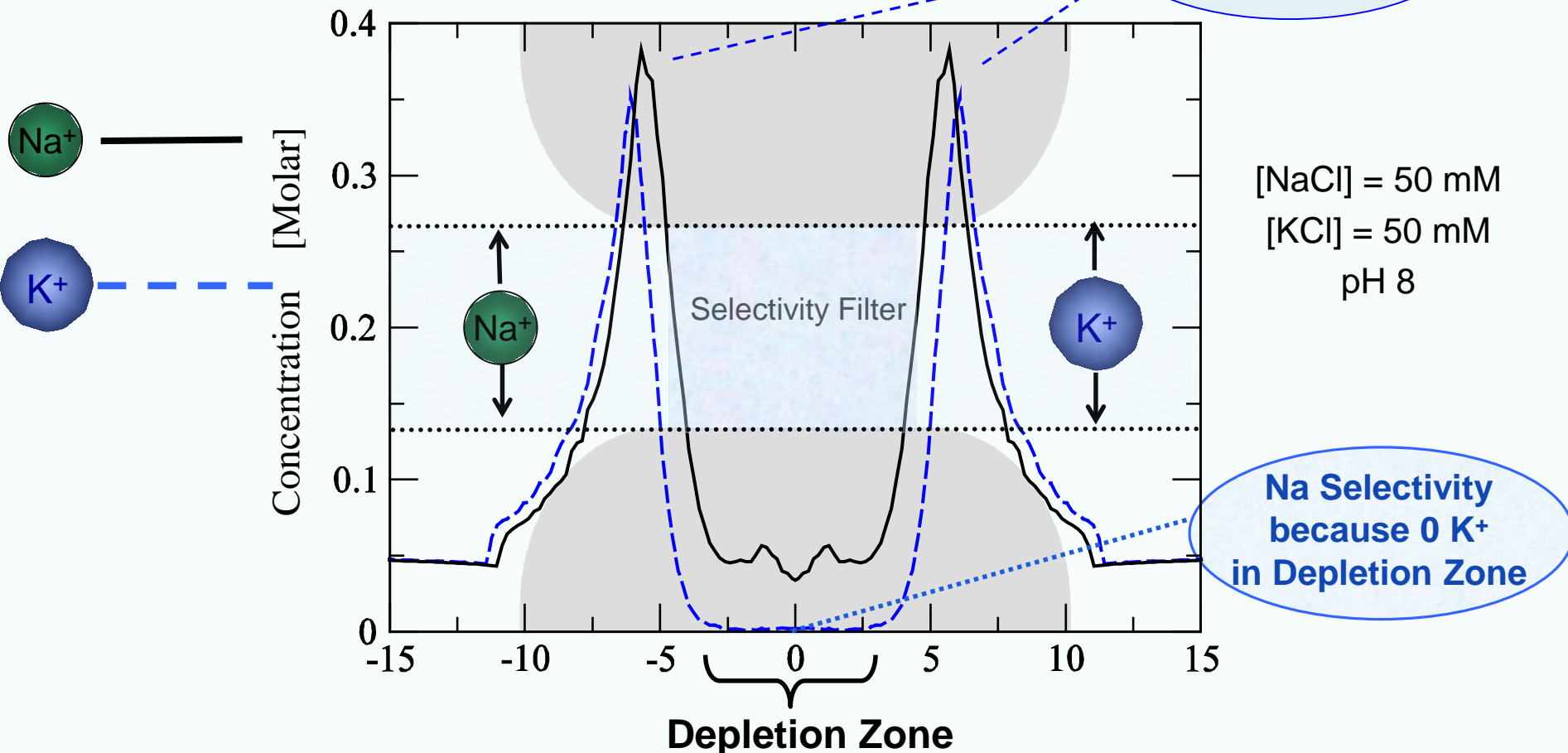
***Nothing was Changed
from the
EEEE Ca channel
except the amino acids***

**Calculated DEKA Na Channel
Selects**

Ca²⁺ vs. Na⁺ and also K⁺ vs. Na⁺

Size Selectivity is in the Depletion Zone

Na⁺ vs. K⁺ Occupancy



of the DEKA Na Channel, 6 Å

How?

Usually Complex Unsatisfying Answers*

How does a Channel Select Na^+ vs. K^+ ?

- * Gillespie, D., Energetics of divalent selectivity in the ryanodine receptor.
Biophys J (2008). 94: p. 1169-1184
- * Boda, *et al*, Analyzing free-energy by Widom's particle insertion method.
J Chem Phys (2011) 134: p. 055102-14

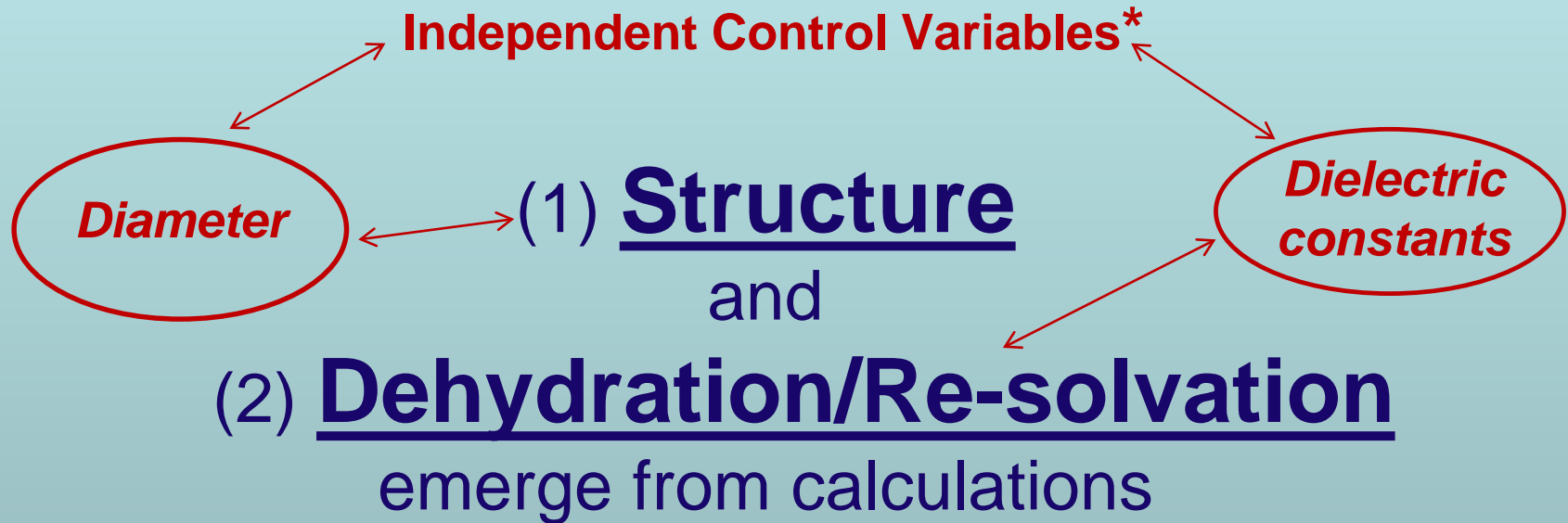
Simple Independent[§] Control Variables*

DEKA Na⁺ channel

*Amazingly simple, not complex
for the most important selectivity property
of DEKA Na⁺ channels*

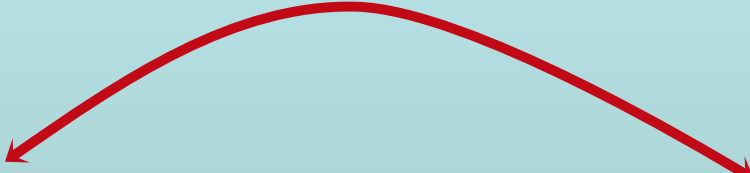
***Control variable** = position of **gas pedal** or dimmer on **light switch**

[§] **Gas pedal** and **brake pedal** are (hopefully) **independent control variables**




Structure (diameter) controls **Selectivity**
Solvation (dielectric) controls **Contents**

**Control variables emerge as outputs
Control variables are not inputs*



Structure (diameter) controls **Selectivity**

Solvation (dielectric) controls **Contents**

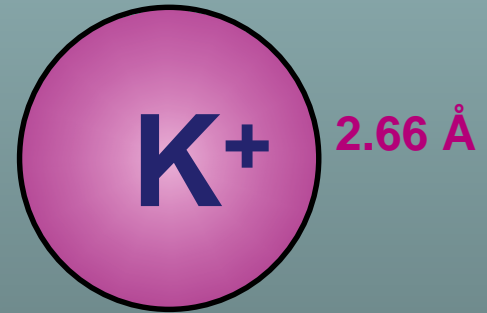
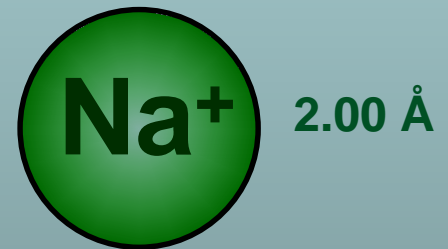
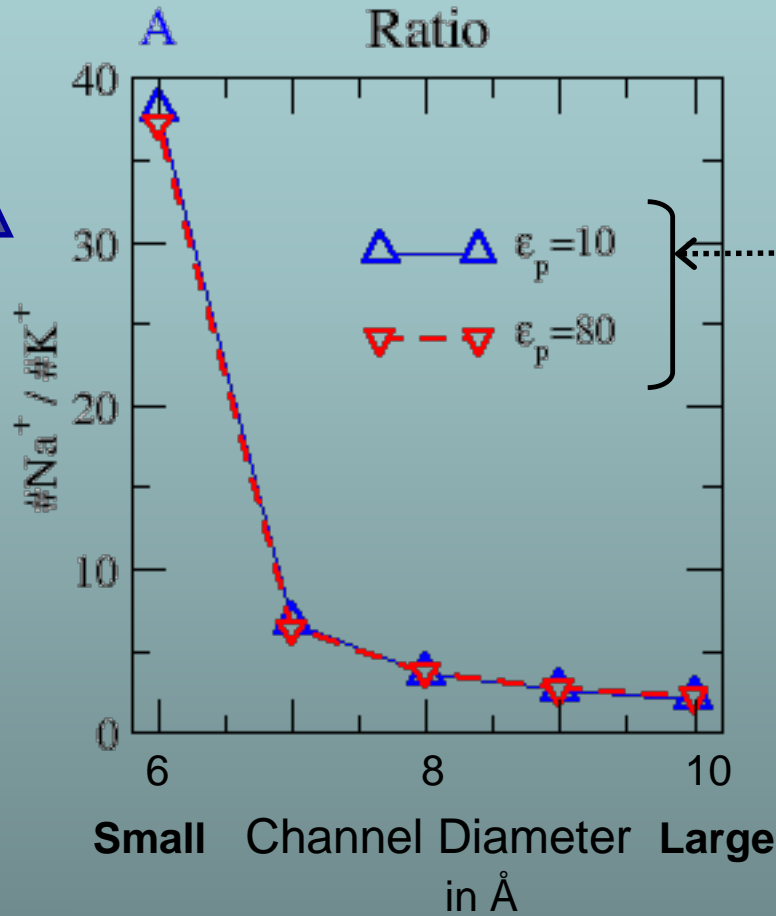
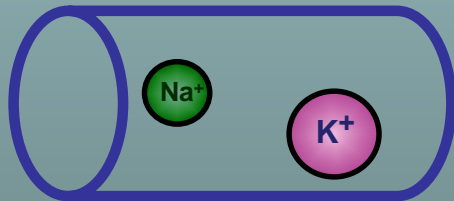


Control Variables emerge as outputs
Control Variables are not inputs

Monte Carlo calculations of the DEKA Na channel

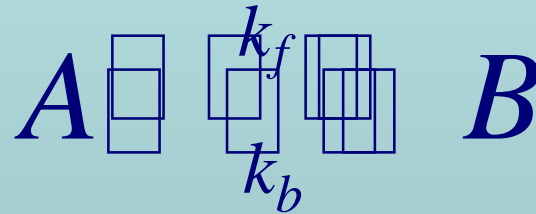
Na⁺ vs K⁺ (size) **Selectivity** (ratio) Depends on **Channel Size**, *not dehydration* (not on Protein Dielectric Coefficient)*

Selectivity
 for small ion



**Generalization
of
'Law' of Mass Action
is needed because
the 'law' assumes
NO FLOW
and
NO INTERACTIONS**

Law of Mass Action



$$-\frac{d}{dt}[A] = k_f [A]; \quad -\frac{d}{dt}[B] = k_b [B]$$

$$K_{eq} = \frac{[B]}{[A]} = \frac{k_f}{k_b}$$

$$J_{A \rightarrow B} = k_f [A]; \quad J_{B \rightarrow A} = k_b [B]$$

[A] means the concentration of species A, i.e., the number density of A

In the minds of most biochemists, many chemists, and textbook authors

A, B are assumed to be **ideal solutions of noninteracting particles**

k_f, k_b are assumed to be **constants independent of concentration** of any species

but

**Everything interacts with
Everything Else**

in biology

and

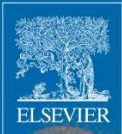
flows cease only at death

Everything is hidden

in K_{eq} , k_f and k_b

*Interactions are significant in biological solutions (Ringer)
Interactions are large in and near channels*

*$k_{f,b}$ and K_{eq} are functions of 'everything'
They are not constants*



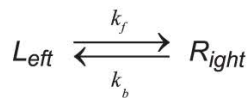
CHEMICAL PHYSICS LETTERS

Editors:
DAVID CLARY
MITCHIO OKUMURA
VILLY SUNDRÖM

Frontiers Editor:
RICHARD SAYKALLY

Frontier research in molecular sciences,
materials and biological systems

'Law' of Mass Action including Flow and Interactions



$$J_k = \underbrace{C_k(L)}_{\text{Source Concentration}} \underbrace{\left(\frac{D_k}{l}\right)}_{\text{Diffusion Velocity}} \underbrace{\text{Prob}\{R|L\}}_{\text{Conditional Probability}} - \underbrace{C_k(R)}_{\text{Source Concentration}} \underbrace{\left(\frac{D_k}{l}\right)}_{\text{Length}} \underbrace{\text{Prob}\{L|R\}}_{\text{Conditional Probability}}$$

Derived from theory of Stochastic Processes

from Bob Eisenberg p. 1-6, in this issue

05.037

Great Opportunity
for
New Science

Chemical Reactions
in
Complex Fluids

Variational Approach
EnVarA

Conservative Dissipative

$$\overbrace{\frac{\delta E}{\delta \mathbf{x}}} - \overbrace{\frac{1}{2} \frac{\delta \Delta}{\delta \mathbf{u}}} = 0$$

Energetic Variational Approach allows

accurate computation of

Flow and Interactions

in Complex Fluids like Liquid Crystals

**Classical theories and Molecular Dynamics
have difficulties with flow, interactions,
and complex fluids**

Engineering needs Calibrated Theories and Simulations

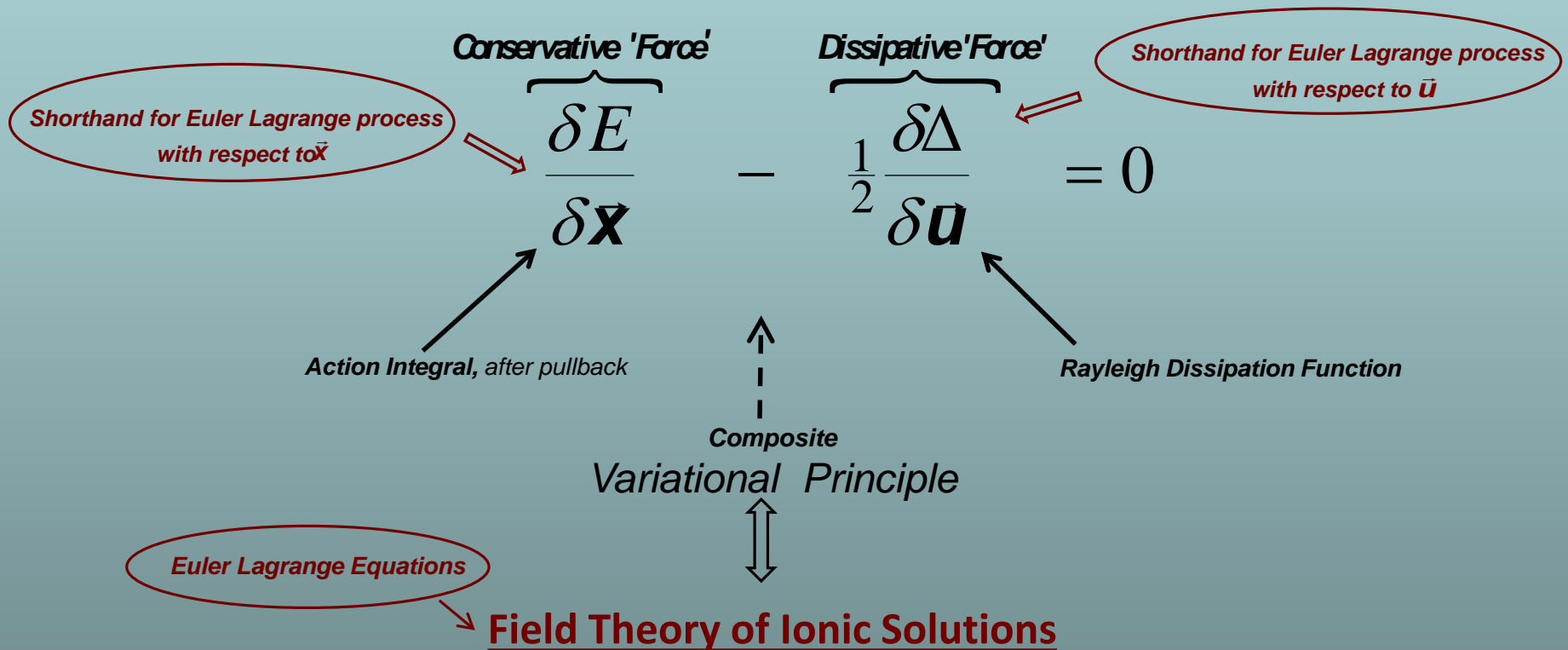
Engineering Devices almost always use flow

Energetic Variational Approach

EnVarA

Chun Liu, Rolf Ryham, Yunkyong Hyon, and Bob Eisenberg

Mathematicians and Modelers: two different 'partial' variations written in one framework, using a 'pullback' of the action integral



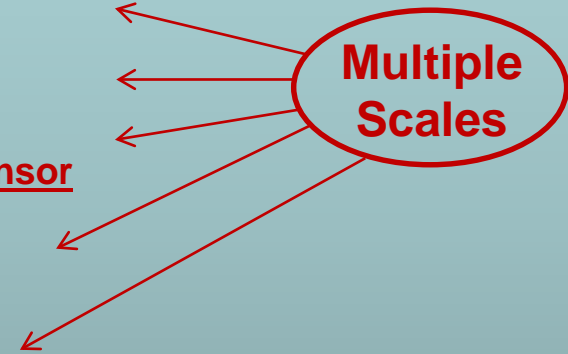
that allows boundary conditions and flow and deals
Consistently with Interactions of Components

Energetic Variational Approach

EnVarA across biological scales: molecules, cells, tissues

Variational theory of complex fluids developed by Chun Liu
with

- | | | | |
|--------------------------------|------------------------|---|-----------------------|
| (1) Hyon, Eisenberg | Ions in | → | <u>Channels</u> |
| (2) Horng, Lin, Liu, Eisenberg | Ions in | → | <u>Channels</u> |
| (3) Bezanilla, Hyon, Eisenberg | Conformation Change of | | <u>Voltage Sensor</u> |
| (4) Ryham, Cohen | Membrane flow | → | <u>Cells</u> |
| (5) Mori, Eisenberg | Water flow in | → | <u>Tissues</u> |



creates a new

Multiscale Field Theory of Interacting Components

needed for Molecular Engineering in general
that allows boundary conditions and flow
and deals with
Ions in solutions self-consistently

Take Home Lessons

Take Home Lesson 1

Ionic Solutions are Complex Fluids

and
cannot be well analyzed by the theory of simple fluids

Take Home Lesson 2

Energetic Variational Approach allows accurate computation of Flow and Interactions in Complex Fluids like Liquid Crystals

Engineering needs Calibrated Theories and Simulations

Engineering Devices almost always use flow

**Classical theories
and
Molecular Dynamics
have difficulties
with
Flow, Interactions,
and Complex Fluids**

Engineering needs Calibrated Theories and Simulations

Engineering Devices almost always use flow

Take Home Lesson 3

Structure
is the
Computed Consequence
of Forces
in these models

Selectivity Depends on Structure

What does the protein do?

Channel and Contents
form a

Self-Organized Structure

with Side Chains at position of
Minimum Free Energy

Protein Fits the Substrate

‘Induced Fit Model of Selectivity’

Binding Sites* are **outputs**
of our Calculations

This is the
**Self-organized
Induced fit model**
of Koshland
and the founders of enzymology,

Made specific by
Mathematics and Computation

Miracle

**We can actually compute the
Structures that determine Selectivity**

New Miracle???

Can *EnVarA* actually compute the function of these systems?

Can *EnVarA* serve as a useful Mathematical Framework for Multi-scale Engineering?

The End

Any Questions?

Solved with Metropolis Monte Carlo

MMC Simulates Location of Ions
both the mean and the variance

Produces Equilibrium Distribution
of location
of Ions and 'Side Chains'

MMC yields Boltzmann Distribution with correct Energy, Entropy and Free Energy

Other methods
give nearly identical results

DFT (Density Functional Theory of fluids, *not electrons*)

DFT-PNP (Poisson Nernst Planck)

MSA (Mean Spherical Approximation)

SPM (Primitive Solvent Model)

EnVarA (Energy Variational Approach)

Non-equil MMC (Boda, Gillespie) several forms

Steric PNP (simplified EnVarA)

Poisson Fermi

Crowded Channels, Crowded Active Sites

are

Complex Fluids

like liquid crystals of LCD displays

All atom simulations of complex fluid are particularly challenging because 'Everything' interacts with 'everything' else on atomic & macroscopic scales

**Generalization
of
“Law of Mass Action”
is
Feasible for
Ionic Solutions**

using the Implicit Solvent Model
of ionic solutions*

*and perhaps the solvent primitive model or more sophisticated models and simulations that professional physical chemists know better than I

Energetic Variational Approach allows

accurate computation of

Flow and Interactions

in Complex Fluids like Liquid Crystals

**Classical theories and Molecular Dynamics
have difficulties with flow, interactions,
and complex fluids**

Engineering needs Calibrated Theories and Simulations

Engineering Devices almost always use flow